

1 **Free Ammonia-Based Sludge Treatment Reduces Sludge Production in the**  
2 **Wastewater Treatment Process**

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14 **Abstract:** Excessive sludge production is one of the major challenges for biological wastewater treatment  
15 plants. This paper reports a new strategy to enhance sludge reduction from the wastewater treatment process.  
16 In this strategy, 1/5 of the sludge is withdrawn from the mainstream reactor into a side-stream unit for sludge  
17 treatment with 16 mg/L free ammonia (FA) for 24-40 h. The FA-treated sludge mixture is then returned to  
18 the mainstream reactor. To demonstrate this concept, two reactors treating synthetic domestic wastewater  
19 were operated, with one serving as the experimental reactor and the other as the control. Experimental  
20 results showed that the experimental reactor exhibited 20% lower in sludge production than the control. FA  
21 treatment effectively disintegrated a portion of extracellular or intracellular substances of sludge cells in the  
22 FA treatment unit and lowered the observed sludge yields in the mainstream reactor, which were the main  
23 reasons for the sludge reduction. Although FA treatment decreased the activities of nitrifiers, denitrifiers,  
24 and polyphosphate accumulating organisms in the FA treatment unit, this strategy did not negatively affect the  
25 reactor performance and sludge properties of the experimental reactor such as sludge settleability, organic  
26 removal, nitrogen removal and phosphorus removal. Further investigation showed that the organics released  
27 from the FA treatment process could be used by PAOs and denitrifiers for carbon sources.

28 **Keywords:** wastewater treatment plants, sludge reduction, biological nutrient removal, free ammonia

## 29 1. Introduction

30 Wastewater treatment plants (WWTPs) are extensively used for biological wastewater treatment to  
31 protect natural water bodies from pollution (Chen et al., 2016; Li et al., 2015; Wang et al., 2012a; 2012b;  
32 2017a). Despite their effective in environmental protection, they are often upset by excessive sludge  
33 production (Li et al., 2016; Hu et al., 2018; Wang et al., 2017b; 2017c; 2018a). For instance, it was reported  
34 that ~34 million metric tons of sludge (~20% of solids) were produced in China in 2015 (Feng et al., 2015; Xu  
35 et al., 2017). The treatment and disposal of excessive sludge is costly, accounting for up to 60% of the total  
36 operational cost of a WWTP (Appels et al., 2008; Wang et al., 2013a; Zhao et al., 2017; Xu et al., 2018). To  
37 enhance sludge reduction, several sludge reduction technologies such as anaerobic fermentation and anaerobic  
38 digestion were successfully developed and extensively implemented in the sludge treatment line in the past  
39 decades (Li et al., 2011; Appels et al., 2008; Rittmann et al., 2008; Wang et al., 2015a; 2015b).

40 Reducing sludge production in the wastewater treatment process (i.e., sludge return line) also attracts  
41 much attention, because this method allows sludge reduction in the first place (i.e., source reduction) and  
42 decreases the costs of subsequent sludge management (Mahmood and Elliott 2006; Wang et al., 2013b).  
43 Besides, sludge contains high levels of organic matters, thus the organics released from sludge reduction in the  
44 sludge return line are returned and reused in microbial metabolisms in the wastewater treatment process.

45 Several sludge treatment methods such as thermal, mechanical, and chemical treatments have been tested  
46 and implemented in the sludge return line to disrupt the extracellular polymeric substances and cell envelopes  
47 (Saby et al., 2002; Dytczak et al., 2007; Wang et al., 2013b). Among them, free nitrous acid (FNA) based  
48 technology seems to be promising, as free nitrous acid is a waste-generated, renewable chemical that can be  
49 produced in situ in WWTPs by side-stream nitrification of the anaerobic digestion liquor (Law et al., 2015;  
50 Wang et al., 2016a; 2016b; 2017d). Wang et al. (2013b) treated 50% of the excess sludge with 2.0 mg/L  
51 FNA for 24-42 h, achieving a 28% reduction in sludge production. Compared with untreated sludge, the  
52 internal organics, denitrification efficiency, and sludge reduction were demonstrated to be increased by 50%,  
53 76%, and 88%, respectively, in a simultaneous fermentation and denitrification reactor fed with sludge  
54 pretreated by FNA (2.04 mg/L) for 24 h (Ma et al., 2015). Despite its effective, this free nitrous acid based  
55 technology requires a side-stream nitrification reactor implemented in WWTPs to produce free nitrous acid.  
56 Unfortunately, the side-stream nitrification reactor is generally non-existent in most of the current WWTPs.

57 Free ammonia (FA), the unionized form of ammonium (i.e.,  $\text{NH}_3$ ), can be obtained up to ~500 mg  
58  $\text{NH}_3\text{-N/L}$  directly from anaerobic digester effluent, which usually contains 1.0-2.0 g  $\text{NH}_4^+\text{-N/L}$  at a pH of  
59 7.5-8.6 and 33 °C (Cervantes. 2009; Fux et al., 2006). Recently, Wei et al. (2017a) found that the sludge  
60 biodegradability was enhanced after FA treatment at 420-680 mg  $\text{NH}_3\text{-N/L}$  for 24 h. Based on the findings,  
61 it is assumed that sludge reduction might be also achieved using sludge treatment by FA through incorporating  
62 a FA treatment unit in the sludge return line to treat part of the return sludge. Unlike free nitrous acid based  
63 technology, this FA method does not require the side-stream nitrification reactor, which thereby has significant  
64 benefits in real-world applications. To date, however, this hypothesis has never been tested.

65 The aim of this work is to evaluate whether this FA-based technology can reduce sludge production and  
66 to assess whether this strategy affect the performance of wastewater treatment. Two bench scale  
67 anaerobic-low dissolved oxygen (DO) sequencing batch reactors receiving synthetic domestic wastewater  
68 were operated to demonstrate the feasibilities. One was carried out as the control, and the other included a  
69 side-stream FA treatment unit was performed as the experimental reactor. When the experimental reactor  
70 achieved in stable operation, FA treatment was implemented through treating 1/5 of the sludge with 16 mg/L  
71 FA for 24-40 h. After steady-state operation, sludge reduction between the two reactors was first assessed  
72 and compared. Then, details of how FA-based sludge treatment enhances sludge reduction were explored.  
73 Finally, the potential impact of FA treatment on the performance of wastewater treatment (e.g., sludge  
74 settleability, organic carbon removal and nutrient removal) was also assessed.

## 75 **2. Materials and methods**

### 76 *2.1. The composition of synthetic wastewater*

77 The synthetic wastewater was used in this study. The synthetic wastewater, unless otherwise described,  
78 contained 320 mg  $\text{CH}_3\text{COONa/L}$ , 136 mg  $\text{NH}_4\text{Cl/L}$ , and 22 mg  $\text{KH}_2\text{PO}_4\text{/L}$  yielding an influent chemical  
79 oxygen demand (COD),  $\text{NH}_4^+\text{-N}$ , and  $\text{PO}_4^{3-}\text{-P}$  of approximately 250, 35, and 5 mg/L, respectively. These  
80 values are close to those measured in the real wastewaters in Central South, China. The concentrations of  
81 other nutrients in the synthetic wastewater were prepared as below (per liter): 0.01 g  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , 0.005 g  
82  $\text{CaCl}_2$ , and 0.5mL of a trace metal solution. The trace metal solution was detailed in our previous publication  
83 (Wang et al., 2008).

### 84 *2.2. The operations of the control and experimental reactors*

85 Two replicate sequencing batch reactors with a working volume of 10 L each were operated and  
86 maintained at  $25 \pm 1$  °C in a temperature controlled room. One reactor was set as the control while the other  
87 was selected as the experimental reactor to examine sludge reduction and nutrient removal by introducing an  
88 FA treatment unit (Figure 1). Both the reactors were carried out with three cycles daily. Each 8 h cycle of  
89 the control reactor (or the experimental reactor) consisted of approximately 120 min anaerobic and 180 min  
90 low DO (0.2-0.5 mg/L) periods, followed by 55 min settling, 5 min decanting, and 120 min idle periods.  
91 Waste activated sludge, which was withdrawn from the secondary sedimentation tank of a municipal  
92 wastewater treatment plant in Changsha, China, was used as the inocula for both two systems.

93 In the decanting period, 6 L of the supernatant was discharged from both reactors and replaced with 6 L  
94 of the synthetic wastewater in the initial 10 min of the anaerobic period. In the low DO period, air was  
95 intermittently supplied into the two reactors through using on/off control systems with online DO detectors to  
96 maintain the DO level between 0.2 and 0.5 mg/L. The reactors were constantly mixed with magnetic stirrers  
97 except for the settling, decanting, and idle periods. The hydraulic retention time (HRT) and sludge retention  
98 time (SRT) in both reactors were the same and maintained at approximately 13.3 h and 20 d, respectively.  
99 The two reactors were operated for about 5 months, which were divided into two phases:

100 Phase I (Baseline phase: Day 0-50): In Phase I, FA treatment was not implemented in the experimental  
101 reactor. The two reactors were operated the same regime as described above for about 50 d to achieve steady  
102 state. SRT in both reactors were maintained at 20 d.

103 Phase II (Experimental phase: Day 51-150): The control reactor in Phase II was operated the same as in  
104 Phase I. Sludge treatment by FA was implemented in the experimental reactor. Approximately 2 L sludge  
105 mixture was daily withdrawn from the experimental reactor at the end of the aerobic period and thickened to  
106 400 ml by centrifugation before being transferred to an FA treatment unit. The pH and  $\text{NH}_4^+\text{-N}$  in the FA  
107 treatment unit were controlled at 8.5 and 80 mg/L. This gives a calculated FA concentration of 16 mg N/L  
108 ( $T = 25$  °C) according to the formula  $\{1.214 \times [\text{NH}_4^+\text{-N}] \times 10^{\text{pH}}\} / \{\exp[6344/(273 + T)] + 10^{\text{pH}}\}$  (Anthonisen  
109 et al., 1976). After 24 h treatment, the FA-treated sludge was transferred to a sludge storage unit on a daily  
110 basis and recirculated to the experimental reactor evenly three times per day in the feeding period. This  
111 actually led to various exposure time of sludge mixture to FA (i.e., 24, 32, and 40 h) for the sludge returned to  
112 the experimental reactor in the three consecutive cycles.

113 2.3. *Batch tests to examine whether the organics released in the FA treatment unit can be used by PAOs and*  
114 *denitrifiers*

115 To examine whether the organics released in the FA treatment unit can be used by PAOs and denitrifiers,  
116 two batch tests (namely, PAOs-Test and Denitrifiers-Test) were operated.

117 PAOs-Test: One batch reactor was operated with a working volume of 500 mL. One 200 mL mixed  
118 liquor was taken from the experimental reactor at the end of the low DO period after stable operation in Phase  
119 II, centrifuged, washed and re-suspended in tap water with a final volume of 200 mL before being moved to  
120 the reactor. Then, 100 mL of FA-treated sludge mixture was centrifuged (5000 rpm) for 5 min before its  
121 supernatant being decanting into the reactor. No additional carbon source was added in batch reactor and pH  
122 in the reactor was maintained ~8.0 manually through dosing 0.5 M HCl or 0.5 M NaHCO<sub>3</sub>. The reactor was  
123 operated under anaerobic condition for 90 min through flushing N<sub>2</sub> into the sludge mixture. Samples were  
124 taken at 30 min intervals for COD, PO<sub>4</sub><sup>3-</sup>-P, polyhydroxyalkanoates (PHA), and glycogen analysis.

125 Denitrifiers-Test: One batch reactor (working volume of 500 mL) was operated and also received the  
126 same amounts of biomass withdrawn from the experimental reactor and supernatant of FA-treated sludge  
127 mixture, as described above. Afterwards, NO<sub>3</sub>-N stock solution was added into the batch reactor, resulting in  
128 an initial NO<sub>3</sub>-N concentration of 35 mg NO<sub>3</sub>-N/L. The reactor was carried out under anoxic condition for  
129 90 min. All other operations were the same as depicted in PAOs-Test. Samples were taken at 15 min  
130 intervals for COD, NO<sub>3</sub>-N, and NO<sub>2</sub>-N analysis.

131 2.4. *Determination of the activities of nitrifiers, denitrifiers, and PAOs*

132 The activities of nitrifiers were expressed by the maximum activities of ammonium-oxidizing bacteria  
133 (AOB) and nitrite-oxidizing bacteria (NOB). For the measurement of the activities of nitrifiers, the  
134 following batch test was carried out. 200 mL sludge mixed liquor was taken from either the experimental  
135 reactor or FA treatment unit, centrifuged (5000 rpm for 5 min), and washed three times with tap water to get  
136 rid of the residual NH<sub>4</sub><sup>+</sup>-N, NO<sub>3</sub>-N and NO<sub>2</sub>-N. Then, the biomass was re-suspended in tap water with a  
137 final volume of 200 mL before being evenly divided into two batch reactors. After that, the two reactors  
138 received either NH<sub>4</sub><sup>+</sup>-N or NO<sub>2</sub>-N stock solution (both at 5.0 g N/L), resulting in an NH<sub>4</sub><sup>+</sup>-N (or NO<sub>2</sub>-N)  
139 concentration of 35 mg NH<sub>4</sub><sup>+</sup>-N/L (or 35 mg NO<sub>2</sub>-N/L). During the whole experiment process, air was  
140 adequately supplied to ensure that DO was not limiting (DO > 5 mg/L). pH in the two reactor were

141 maintained at  $\sim 7.5$  by addition of 0.5 M HCl or 0.5 M NaHCO<sub>3</sub> solution. The test was lasted for 60 min.  
142 Samples were taken at 15 min intervals for the analyses of NH<sub>4</sub><sup>+</sup>-N and NO<sub>3</sub><sup>-</sup>-N. The MLVSS concentration  
143 in both reactors was measured when the experiment was finished. The activities of AOB and NOB were  
144 respectively determined by the biomass-specific NH<sub>4</sub><sup>+</sup>-N consumption and NO<sub>3</sub><sup>-</sup>-N production rates, which  
145 were evaluated by dividing the corresponding volumetric rates by the MLVSS concentration.

146 For the assessment of the activities of denitrifiers, the following batch experiment was performed. A  
147 mixed liquor sample (100 mL) was taken out from either the experimental reactor or FA treatment unit,  
148 centrifuged, washed and re-suspended in tap water with a final volume of 100 mL before being transferred in  
149 the batch reactor. N<sub>2</sub> was flushed through the sludge to get rid of any residual oxygen. CH<sub>3</sub>COONa and  
150 NO<sub>3</sub><sup>-</sup>-N stock solution were added into the reactor, resulting in initial COD and NO<sub>3</sub><sup>-</sup>-N concentrations of 500  
151 mg COD/L and 100 mg NO<sub>3</sub><sup>-</sup>-N /L, respectively. pH in the reactor was controlled at  $\sim 7.5$  by dosing 0.5 M  
152 HCl or 0.5 M NaHCO<sub>3</sub> solution. The tests lasted for 60 min. Mixed liquor samples were taken regularly  
153 (15 min intervals) for NO<sub>3</sub><sup>-</sup>-N analysis. The MLVSS concentrations were measured at the end of the test.  
154 The activities of denitrifiers were determined as the biomass-specific NO<sub>3</sub><sup>-</sup>-N reduction rate, which was  
155 calculated by dividing the volumetric NO<sub>3</sub><sup>-</sup>-N reduction rate by the MLVSS concentration.

156 For the assessment of the activity of PAOs, a mixture (100 mL) was taken out from either the  
157 experimental reactor or FA treatment unit, centrifuged, washed and re-suspended in tap water with a final  
158 volume of 100 mL before being transferred in one batch reactor. Then, the COD concentration in the liquid  
159 of the reactor was controlled at 500 mg/L (prepared by CH<sub>3</sub>COONa). pH in the reactor was maintained  
160  $\sim 8.0$  manually through dosing 0.5 M HCl or 0.5 M NaHCO<sub>3</sub>. The reactor was operated as the anaerobic (2  
161 h)-low DO (0.2~0.5 mg/L, 3h) regime. Samples were taken at 30 min intervals for PO<sub>4</sub><sup>3-</sup>-P analysis. The  
162 activity of PAOs was determined by the biomass-specific PO<sub>4</sub><sup>3-</sup>-P release or uptake rate.

### 163 2.5. Sampling and analytical methods

164 The mixed liquor volatile solid (MLSS) and MLVSS concentrations in the two reactors were measured  
165 twice a week in Phase I and once a week in Phases II. Sludge reduction was determined by comparing the  
166 quantity of sludge produced from the control and experimental reactors. The effluent NH<sub>4</sub><sup>+</sup>-N, NO<sub>3</sub><sup>-</sup>-N, NO<sub>2</sub><sup>-</sup>  
167 -N, and COD concentrations were measured twice a week. SVI (sludge volume index, twice a week) and  
168 particle size distribution in sludge (once a week) were also determined.

169 All samples were taken using a pipette and immediately filtered through a Whatman GF/C glass  
170 microfiber filter (0.22  $\mu\text{m}$  pore size) for the measurements of  $\text{NH}_4^+\text{-N}$ ,  $\text{NO}_3^-\text{-N}$ ,  $\text{NO}_2^-\text{-N}$ ,  $\text{PO}_4^{3-}\text{-P}$  and COD in  
171 accord with standard methods (APHA, 1998). The MLSS and MLVSS concentrations as well as SVI were  
172 determined according to the standard methods (APHA, 1998). Particle size distribution analysis was  
173 performed using a Malvern Mastersizer 2000 instrument (Malvern Instruments, Worcestershire, UK) with a  
174 detection range of 0.01 ~ 3500  $\mu\text{m}$ . The observed sludge yield ( $Y_{\text{obs}}$ : mg MLSS/mg COD) was calculate by  
175 sludge production rate to the COD removed in the reactor (Metcalf and Eddy, 2003). The determination of  
176 PHA (calculated as the sum of poly-3-hydroxybutyrate, poly-3-hydroxyvalerate, and  
177 poly-3-hydroxy-2-methylvalerate) and glycogen was conducted the same as described prior studies (Wang et  
178 al., 2009). Excitation emission matrix (EEM) fluorescence spectroscopy ((Fluoromax-4 Spectrofluorometer,  
179 HORIBA Scientific, France) with a 450 W Xe arc lamp) was utilized to characterize the changes of FA-treated  
180 sludge liquid according to the method reported previously (Luo et al., 2013). Fluorescence in situ  
181 hybridization was used to quantify the abundances of PAOs, AOB, and NOB in the two reactors, and the  
182 procedures were detailed prior studies (Wang et al., 2012b; 2016b).

### 183 2.6. Statistical Analysis

184 An analysis of variance was used to evaluate the significance of results, and  $p < 0.05$  ( $p > 0.05$ ) was  
185 considered to be statistically significant (insignificant).

## 186 3. Results and discussion

### 187 3.1. Comparison of sludge production between the two reactors

188 Fig. 2 shows that the concentrations of MLSS and MLVSS in the two reactors during the long-term  
189 operation. It can be seen that both the MLSS and MLVSS concentrations in the control reactor remained  
190 stable over the whole operation period. In Phase I (i.e., before implementing FA treatment), the experimental  
191 reactor had MLSS and MLVSS concentrations similar to those in the control reactor. When FA treatment was  
192 implemented, the MLSS and MLVSS concentrations decreased gradually during the initial two weeks and  
193 then remained stable in the remainder period.

194 To further assess feasibility of FA-based sludge treatment achieves sludge reduction, Table 1 summaries  
195 the concentrations of MLSS and MLVSS as well as sludge production rates in the reactors in stable operation

196 before and after implementing FA treatment. In steady-state operation of Phase I, MLSS and MLVSS in the  
197 two reactors were similar and maintained at ~2200 mg/L and ~1900 mg/L, respectively. When FA treatment  
198 was implemented in the experimental reactor, MLSS and MLVSS concentrations reduced to ~1800 mg/L and  
199 ~1500 mg/L, respectively. Sludge production rate in this reactor decreased from ~1130 to ~920 mg MLSS/d.  
200 According to our calculation, a sludge reduction of approximately 20% was achieved in the experimental  
201 reactor in stable operation of Phase II using FA treatment.

### 202 3.2. Details of how FA treatment causes sludge reduction

203 In order to reveal the details of how FA treatment achieves sludge reduction, we first figured out the fact  
204 of what happens in FA treatment unit. Sludge cells mainly consist of extracellular polymeric substances and  
205 intracellular substances, and their releases could be indicated by EEM fluorescence spectroscopy. Previous  
206 studies confirmed that protein comprises large amount of aromatic structures, which have fluorescence  
207 characteristics. In this work, EEM was used to characterize the changes of sludge liquid before (Fig. 3a) and  
208 after FA treatment (Fig. 3b). The Peak A and Peak B, which are respectively located at the Ex/Em of  
209 275-280/ 335-355 and 220-225/335-345 nm in EEM spectra and related to tryptophan protein-like substances  
210 (peak A) and aromatic proteins (peak B) (Sheng and Yu, 2006; Luo et al., 2013), were observed in both  
211 samples. Compared with the untreated sludge, FA-treated sample had stronger the fluorescence intensities of  
212 the peaks A-B in EEM spectra, indicating that more protein was present in the sludge supernatant liquor after  
213 FA treatment. The results suggested that FA treatment led to either extracellular polymeric substances or  
214 intracellular substances released from solid phase to liquid phase (Wang et al., 2018b; Wei et al., 2017a).

215 Fig. 3c shows the profiles of cumulative floc size distribution of sludge flocs before and after FA  
216 treatment. The sludge with 24h FA treatment performed a shift in particle size distribution towards smaller  
217 sizes with the peak at 79-104  $\mu\text{m}$  than sludge without FA treatment, which had the most volume density  
218 between 181 and 208  $\mu\text{m}$ , suggesting that FA treatment decreased the floc size of sludge. All these results  
219 indicated that FA treatment caused the disruption of sludge flocs. It was reported that FA could break down  
220 the recalcitrant structure of sludge flocs and led to a rapid increase of soluble substrates (Hejnfelt and  
221 Angelidaki, 2009).

222 Table 2 further exhibits the variations of soluble COD,  $\text{NH}_4^+\text{-N}$ , and activities of nitrifiers, denitrifiers,  
223 and PAOs in FA treatment unit before and after FA treatment at steady-state in Phase II. It was found that FA

224 treatment increased the concentrations of soluble COD and  $\text{NH}_4^+\text{-N}$ . After 24 h FA treatment,  $729 \pm 9$  mg/L  
225 of soluble COD and  $90.8 \pm 3.6$  mg/L of  $\text{NH}_4^+\text{-N}$  were measured, respectively. This confirmed again that FA  
226 treatment could enhance the disruption of sludge flocs. Previous studies also verified that the extracellular  
227 materials of sludge cells were disrupted into debris by FA treatment (Wei et al., 2017a; 2017b), which was in  
228 accord with the findings obtained in this work. The direct disruption of sludge flocs was one reason for the  
229 enhanced sludge reduction achieved in the experimental reactor.

230 Also from Table 2, it can be seen that the activities of AOB, NOB, denitrifiers, and PAOs were all  
231 inhibited by FA treatment. The activities of AOB and NOB decreased from  $9.8 \pm 0.6$  and  $8.1 \pm 0.7$  mg N/g  
232 VSS•h to  $7.7 \pm 0.7$  and  $4.3 \pm 0.5$  mg N/g VSS•h after 24 h FA treatment, respectively, suggesting that FA  
233 treatment resulted in a 21% decrease in AOB activity and a 47% decrease in NOB activity. Similar to  
234 nitrifiers, the activity of denitrifiers was inhibited by 36% after 24 h FA treatment (the denitrification rate  
235 decreased from  $95.6 \pm 3.6$  to  $61.2 \pm 3.1$  mg  $\text{NO}_3\text{-N/g VSS h}$ ). However, the activity of PAOs was observed  
236 to be inhibited severely (>90%). After 24 h treatment by FA, anaerobic phosphate release rate decreased  
237 from  $20.9 \pm 1.4$  to  $1.7 \pm 0.4$  mg P/g VSS•h while aerobic phosphate uptake rate decreased from  $14.2 \pm 0.9$  to  
238  $0.8 \pm 0.3$  mg P/g VSS•h. It is well-known that the enzymes of these bacteria are cell membrane-bound  
239 enzymes which are protected by extracellular polymeric substances. When sludge cells especially  
240 extracellular polymeric substances are disrupted by the biocidal effect of FA, these membrane-bound enzymes  
241 are easier to be attacked by FA. Previous investigation showed that FA could diffuse through the cell  
242 membrane, shuttle protons between the two sides without energy consumption, and caused cell inactivation by  
243 disrupting the proton and potassium balance inside the cell (Hejnfelt and Angelidaki, 2009). Possibly due to  
244 this inactivation, a lower  $Y_{\text{obs}}$  (observed sludge yields) was achieved in the experimental reactor at steady  
245 Phase II, as compared with that in the control ( $0.21 \pm 0.01$  vs  $0.25 \pm 0.01$  mg MLSS/mg COD), it can be  
246 understood why lower sludge production was achieved in the experimental reactor.

### 247 3.3. Effect of FA treatment on sludge properties and reactor performance

248 In Phase I (i.e., before implementing FA treatment), similar reactor performance was observed in the  
249 control and experimental reactors (i.e., effluent COD,  $\text{NH}_4^+\text{-N}$ ,  $\text{NO}_3\text{-N}$ ,  $\text{NO}_2\text{-N}$ , and  $\text{PO}_4^{3-}\text{-P}$  concentrations) and  
250 sludge properties (i.e., SVI, particle size, and activities and abundances of AOB, NOB, and PAOs) (please see  
251 Table S1 for details). For instance, 0.5 and 0.4 mg/L of  $\text{PO}_4^{3-}\text{-P}$ , 0.2 and 0.3 mg/L of  $\text{NH}_4^+\text{-N}$ , 0.1 and 0.2

252 mg/L of  $\text{NO}_2\text{-N}$ , and 2.0 and 2.3 mg/L of  $\text{NO}_3\text{-N}$  were detected in the control and experimental reactors,  
253 respectively, suggesting that more than 90% of phosphate removal and 92% of nitrogen removal were  
254 achieved in both reactors.

255 To assess whether FA treatment affects the reactor performance and sludge properties, we compared the  
256 parameters above of the two reactors at steady-state Phase II, and the results were summarized in Table 3. It  
257 was observed that particle size ( $d_{50}$ ) in the experimental reactor was ~10% lower than that in the control  
258 reactor (Table 3 and Fig. 4). As discussed in Section 3.2, FA treatment could disintegrate extracellular  
259 materials and/or the bacterial cells. This is possibly to result in the generation of smaller particles and then  
260 accumulated in the experimental reactor, which might be the reason for the relatively smaller particle sizes in  
261 the experimental reactor. Although FA treatment decreased particle size of sludge cells, sludge settleability  
262 was not affected by this treatment, which was reflected by SVI. It was found that  $134 \pm 5$  and  $142 \pm 7$  mL/g  
263 of SVI were respectively measured in the control and experimental reactors, and this difference was found to  
264 be statistically insignificant ( $p > 0.05$ ).

265 The activities and abundances of AOB and NOB in the two reactors were detected to evaluate if FA  
266 treatment affects nitrifiers. It can be seen that there was no significant change ( $p > 0.05$ ) in terms of the  
267 activities of AOB and NOB between the control and experimental reactor at steady Phase II, suggesting that  
268 the activities of nitrifiers were not affected when FA treatment was incorporated. In addition, the populations  
269 of AOB and NOB in the control reactor were also similar to those in the experimental reactor ( $p > 0.05$ ). The  
270 abundances of AOB and NOB were respectively  $6.5 \pm 0.8\%$  and  $3.3 \pm 0.4\%$  in the control reactor, while the  
271 corresponding data were  $7.4 \pm 1.1\%$  and  $3.1 \pm 0.3\%$ , respectively. Moreover, the activity of denitrifiers was  
272 also similar in the control and experimental reactors ( $97.6 \pm 5.9$  vs  $93.9 \pm 4.7$  mg N/g VSS•h). These results  
273 demonstrated that both nitrification and denitrification in the mainstream reactor was unaffected by FA  
274 treatment. As a result, the two reactors had almost the same  $\text{NH}_4^+\text{-N}$ ,  $\text{NO}_3\text{-N}$ , and  $\text{NO}_2\text{-N}$  in the effluent  
275 (Table 3).

276 As shown in Table 2, the activity of PAOs in the FA treatment unit was severely inhibited by FA for 24 h  
277 (>90%). However, this inhibition was largely mitigated in the mainstream experimental reactor. In stable  
278 operation,  $15.7 \pm 0.7$  mg P/g VSS•h of anaerobic phosphate release rate and  $10.6 \pm 0.4$  mg P/g VSS•h of  
279 aerobic phosphate uptake rate were respectively measured in the experimental reactor, which were 77.3% and

280 75.7% of those determined in the control reactor. The results indicated that the decreased activity of PAOs in  
281 the FA treatment unit was probably due to inhibition rather than inactivation, and the activity of PAOs could  
282 be recovered after several days of recovery time. In the literature, Wang et al. also found that the activities of  
283 AOB and NOB could be recovered to some extents after 4 d when they were treated by free nitrous acid  
284 (Wang et al., 2016b). Further investigation verified that the abundance of PAOs did not decrease largely,  
285 which was consistent with the data of anaerobic phosphate release and aerobic phosphate uptake rates.  
286 Therefore, an excellent phosphate removal was also obtained in the experimental reactor at steady state  
287 operation, which was comparable with that in the control reactor ( $0.6 \pm 0.1$  mg/L vs  $0.4 \pm 0.1$  mg/L of  
288 phosphate in effluent).

#### 289 *3.4. Could the organics released in the FA treatment unit be used by PAOs and denitrifiers*

290 FA treatment caused a large amount of organics released from sludge cells into liquid phase. After 24 h  
291 treatment,  $729 \pm 28$  mg/L soluble COD was measured in the FA treatment unit (Table 2). This soluble COD  
292 increased influent organic loading by ~6% in the experimental reactor. However, no significant increase of  
293 effluent COD was measured in the experimental reactor, as compared with that in the control reactor (Table 3,  
294  $p > 0.05$ ). This suggested that the organics released from sludge flocs could be metabolized in the  
295 experimental reactor. It is known that many biodegradable organics could be degraded under aerobic  
296 conditions by ordinary heterotrophic bacteria through the tricarboxylic acid cycle, however, the question as to  
297 whether the organics released from sludge flocs could be used by PAOs and denitrifiers for supplementary  
298 carbon sources remained unknown.

299 To assess this feasibility, batch tests were operated using the organics released as the sole carbon source,  
300 and the results were outlined in Table 4. It can be seen that soluble COD decrease coupled with anaerobic  
301 phosphate release and PHA accumulation occurred concurrently in PAOs-test. The behavior observed was  
302 similar to the typical phenotype of PAOs. It should be emphasized that  $163 \pm 8$  mg/L of soluble COD was  
303 still determined after 90 min anaerobic period. Moreover, the levels of phosphate release and PHA  
304 accumulation were lower than those reported previously (Wang et al., 2012a; 2013a; Yi et al., 2017). The  
305 results suggested that not all but only a part of organics released could be metabolized by PAOs. The  
306 released organics included 52.3% protein, 19.8% carbohydrate, 5.4% volatile fatty acids, and 22.5% other  
307 organic compounds (Table 2). Previous investigations demonstrated that some of them such as acetate,

308 propionate, glucose, and alcohol could be used by PAOs, and no information was available to the feasibility  
309 that protein could be utilized by PAOs before its degradation to small organics (Oehmen et al., 2007; Yang et  
310 al., 2018; Wang et al., 2008; 2012a; 2013a).

311 In Denitrifiers-Test, both soluble COD and  $\text{NO}_3\text{-N}$  concentrations decreased with the reaction time. At  
312 90 min anoxic time,  $135 \pm 6$  mg/L soluble COD was measured while nitrate was at non-detectable level. The  
313 results clearly verified that the organics released could be used as carbon sources for denitrification. By  
314 comparing the endogenous denitrification rates in the control and experimental reactors, Wang et al. also  
315 demonstrated the availability of free nitrous acid-treated sludge as a carbon source for denitrification (Wang et  
316 al., 2013b).

### 317 3.5. Implications

318 This paper reports for the first time that FA-based sludge treatment is effective in sludge reduction from  
319 the wastewater treatment process. This was experimentally demonstrated by comparing MLSS concentration  
320 and sludge production rates between the two reactors with and without FA treatment. The recirculation of 2  
321 L of the sludge that was treated using FA at 16 mg N/L for 24-40 h to the experimental reactor, 20% lower  
322 sludge production was achieved, as comparing with the control reactor without FA treatment. Meanwhile,  
323 the reactor performances and sludge properties were not affected by this novel FA-based sludge reduction  
324 strategy, suggesting that FA has good potential to be developed into a practical technology in field situations.

325 In the past years, several sludge treatment methods such as thermal, mechanical, and chemical treatments  
326 have been tested and implemented in the sludge return line to enhance sludge reduction (Saby et al., 2002;  
327 Dytczak et al., 2007; Wang et al., 2013b). Despite these effective, these methods require high input of either  
328 chemicals or energy, which diminish their values in real-world applications. In comparison, however, FA is a  
329 waste-generated, renewable chemical that can be produced in situ in WWTPs as a byproduct of sludge  
330 treatment (i.e., FA can be attained directly from anaerobic digester/fermenter effluent), suggesting that the  
331 FA-based sludge treatment method does not have these drawbacks (Wang et al., 2017e). The findings  
332 achieved in this work, therefore, may guide engineers to develop more economic and practical strategies for  
333 sludge reduction from the sludge return line.

334 Fig. 5b presents a “closed-loop” concept in a WWTP for the enhanced sludge reduction from the sludge  
335 return line based on the proposed FA-based sludge treatment technology. Apart from returning a part of

336 sludge, the excess sludge withdrawn from the settler was generally digested anaerobically to reduce the  
337 amount of sludge, kill the pathogenic microorganisms, and produce energy biogas methane (Appels et al.,  
338 2008; Wang et al., 2015b). The digestion liquid, which usually contains 1.0-2.0 g  $\text{NH}_4^+$ -N/L at a pH of  
339 7.5-8.6 and 33 °C (Cervantes. 2009; Fux et al., 2006), requires to be recirculated to the head of the WWTP to  
340 be cleaned before its final discharge (Fig. 4a). By setting up a FA treatment unit in the sludge return line and  
341 introducing the digestion liquid into the FA treatment unit instead of the head of WWTP, this FA-based sludge  
342 reduction strategy can be readily implemented (Fig. 5b).

343 Based on the experimental data obtained and key operating parameters used in this work, an expanded  
344 research on a full-scale WWTP with a 100,000 population equivalent (PE) was used to investigate the  
345 potential economic benefit of this FA-based technology (Please see Table S2 for details). It was estimated  
346 that compared to an extended aeration WWTP, total volume of the reactor (bioreactor + FA treatment reactor)  
347 of a conventional WWTP based on FA treatment would be reduced by 20% with the same sludge production.  
348 This is equivalent to capital investment savings of \$1,171,200, with an annual \$123,500 saving over a 20-year  
349 period by assuming an interest rate of 8.5%. Furthermore, requiring a smaller footprint is crucial and  
350 beneficial for WWTPs under the condition of limited land availability. In terms of energy consumption, the  
351 annual cost of the system with FA treatment (mixing + recirculation) was estimated to be \$50,430 lower  
352 compared to the extended aeration system (mixing). The FA treatment unit is expected to consist of a very  
353 simple vessel and simple mixing devices, which is the opposite of those currently available methods (i.e.  
354 thermal, mechanical and chemical) that require professional vessels and equipment to meet high temperature,  
355 high pressure and high mechanical forces conditions. Moreover, FA is a renewable, cheap chemical that can  
356 be obtained in-situ production in a WWTP. Thus, the cost of FA and FA treatment unit is very low. In  
357 summary, WWTP with the FA method would save \$172,930/year. The FA strategy for achieving sludge  
358 reduction is economically feasible and environmentally friendly.

359 It should be emphasized that this study mainly aimed to assess the feasibility of whether FA treatment  
360 could reduce sludge production, thus technical optimization was not performed. For example, several  
361 parameters such as the FA level, treatment time, and sludge portion for treatment were not optimized. In  
362 addition, one bench-scale experiment was tested in this work. Thus, the benefit and cost values presented  
363 above should be considered indicative only, its technical and economic feasibilities under real-world situations

364 require to be further evaluated under real-world situations.

#### 365 **4. Conclusions**

366 This paper presents a new strategy about sludge reduction in wastewater treatment process using FA  
367 treatment. Experimental results showed that when 2 L sludge was treated with 16 mg N/L FA for 24-40 h  
368 every day, 20% of sludge reduction was achieved in the experimental reactor. FA treatment caused the  
369 release of large amounts of extracellular or intracellular substances in the FA treatment unit and lowered the  
370 observed sludge yields in the mainstream reactor, which were the main reason for the sludge reduction.  
371 Although FA treatment decreased the activities of function bacteria in the FA treatment unit, no adverse  
372 impacts on reactor performance and sludge properties were found in the mainstream reactor. Further  
373 investigation showed that the organics released in the FA treatment process could be used by PAOs and  
374 denitrifiers.

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#### 379 **Supporting Information**

380 This file contains Tables S1-S2.

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487 **Figure legends**488 **Fig. 1.** Schematic diagram of the control (A) and experimental systems (B).489 **Fig. 2.** Long-term MLSS and MLVSS concentrations in the control and experimental reactors. Phase I: before  
490 FA treatment (baseline phase); Phase II: FA treatment implemented in the experimental reactor (experimental  
491 phase).492 **Fig. 3.** EEM profiles of liquid (a: before treatment; b: after treatment) and particle size distribution of sludge  
493 (c) before and after FA treatment in the FA treatment unit.494 **Fig. 4.** Particle size distribution of sludge in the control and experimental treatment reactors.495 **Fig. 5.** Traditional wastewater treatment process (a), innovative “closed-loop” concept in wastewater  
496 treatment process based on the proposed FA treatment technology to achieve sludge reduction (b).

497 **Table legends**498 **Table 1** Summarization of the results associated with sludge reduction.499 **Table 2** Changes of soluble COD,  $\text{NH}_4^+$ -N and activities of nitrifiers, denitrifiers, and PAOs in FA treatment unit.500 **Table 3** Treatment performance and sludge properties of the control and experimental reactors at steady phase II.501 **Table 4** Changes of soluble COD,  $\text{PO}_4^{3-}$ -P, PHA, glycogen in PAOs-Test, and variations of COD, and  $\text{NO}_3^-$ -N in  
502 Denitrifiers-Test.

**Table 1** Summarization of the results associated with sludge reduction <sup>a</sup>

Parameter	Phase I		Phase II	
	Control reactor	Experimental reactor	Control reactor	Experimental reactor
MLSS (mg/L)	2242 ± 21	2256 ± 25	2253 ± 24	1807 ± 18
MLVSS (mg/L)	1896 ± 17	1907 ± 20	1911 ± 19	1505 ± 15
Sludge production rate (mg MLSS/d)	1123 ± 16	1128 ± 15	1126 ± 18	924 ± 16
Sludge reduction	Not applicable		(2253-1807)/2253 =20%	
Y <sub>obs</sub> (mg MLSS/mg COD)	0.24 ± 0.01	0.24 ± 0.01	0.25 ± 0.01	0.21 ± 0.01
MLVSS/MLSS (%)	84.6 ± 0.1	84.2 ± 0.1	85.0 ± 0.2	83.3 ± 0.1

<sup>a</sup> Results are the averages and standard deviations from seven measurements during steady-state in Phase II.

**Table 2** Changes of soluble COD, NH<sub>4</sub><sup>+</sup>-N and activities of nitrifiers, denitrifiers, and PAOs in FA treatment unit <sup>a</sup>

Treatment time	Soluble COD (mg/L)	NH <sub>4</sub> <sup>+</sup> -N (mg/L)	AOB activity (mg N/g VSS·h)	NOB activity (mg N/g VSS·h)	P release rate (mg P/g VSS·h)	P uptake rate (mg P/g VSS·h)	Denitrification rate (mg NO <sub>3</sub> <sup>-</sup> -N/g VSS h)
0	15 ± 3	0.2 ± 0.1	9.8 ± 0.6	8.1 ± 0.7	20.9 ± 1.4	14.2 ± 0.9	95.6 ± 3.6
24 h	729 ± 28 <sup>b</sup>	90.8 ± 3.6	7.7 ± 0.7	4.3 ± 0.5	1.7 ± 0.4	0.8 ± 0.3	61.2 ± 3.1

<sup>a</sup> Results are the averages and standard deviations from triplicate tests during steady-state in Phase II. The pH in the FA treatment unit was 8.5 during the entire treatment period. <sup>b</sup> The soluble COD included 52.3% protein, 19.8% carbohydrate, 5.4% volatile fatty acids, and 22.5% others.

**Table 3** Treatment performance and sludge properties of the control and experimental reactors at steady phase II <sup>a</sup>

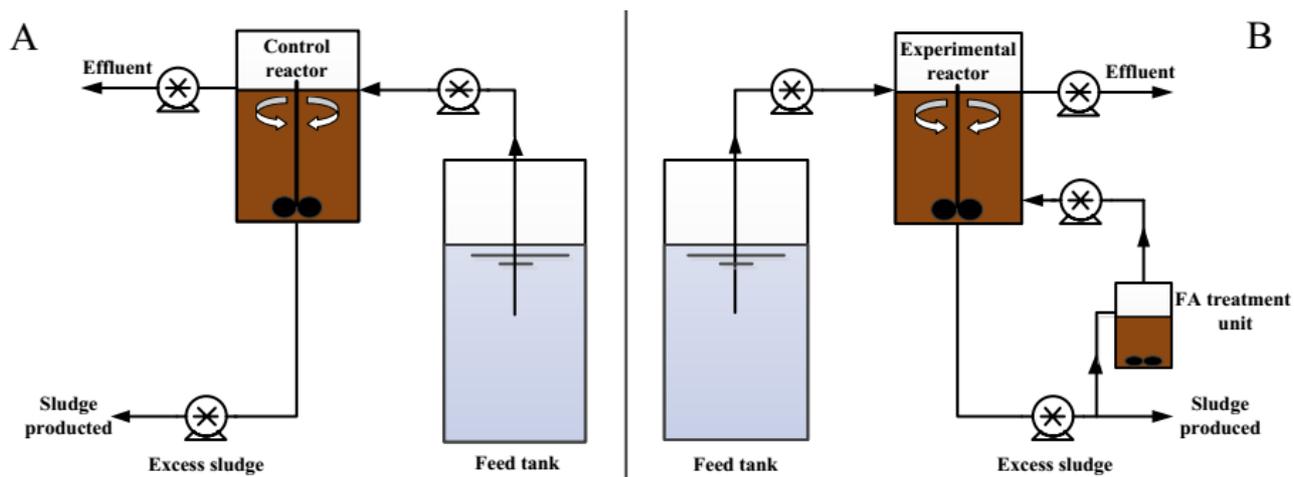
Parameter		Control reactor	Experimental reactor
Treatment performance	Effluent COD (mg/L)	13.9 ± 4.2	16.6 ± 3.7
	Effluent PO <sub>4</sub> <sup>3-</sup> (mg/L)	0.4 ± 0.1	0.6 ± 0.1
	Effluent NH <sub>4</sub> <sup>+</sup> concentration (mg N/L)	0.2 ± 0.1	0.1 ± 0.1
	Effluent NO <sub>2</sub> concentration (mg N/L)	0.1 ± 0.1	0.1 ± 0.1
	Effluent NO <sub>3</sub> concentration (mg N/L)	2.2 ± 0.5	2.5 ± 0.6
Sludge properties	SVI (mL/g)	134 ± 5	142 ± 7
	Particle size (d50) <sup>b</sup> (μm)	206 ± 7	183 ± 10
	AOB abundance (%)	6.5 ± 0.8	7.4 ± 1.1
	AOB activity (mg N/g VSS·h)	7.94 ± 0.12	8.05 ± 0.17
	NOB abundance (%)	3.3 ± 0.4	3.1 ± 0.3
	NOB activity (mg N/g VSS·h)	7.15 ± 0.11	6.52 ± 0.14
	PAO abundance (%)	17.7 ± 3.2	15.6 ± 2.7
	P release rate (mg P/g VSS·h)	20.3 ± 0.5	15.7 ± 0.7
	P uptake rate (mg P/g VSS·h)	14.0 ± 0.3	10.6 ± 0.4
	Denitrifiers' activity (mg N/g VSS·h)	97.6 ± 5.9	93.9 ± 4.7

<sup>a</sup> Results are the averages and standard deviations from triplicate measurements during steady-state in Phase II. <sup>b</sup> d50 denotes the size for which 50% (measured based on volume) of particles in the sample is below this size.

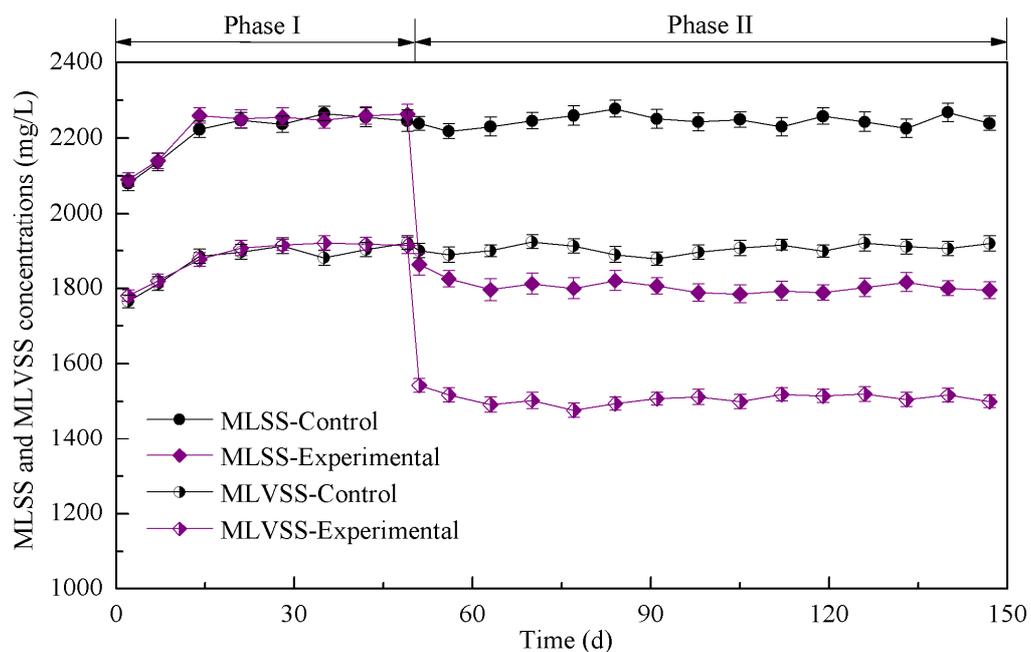
**Table 4** Changes of soluble COD, PO<sub>4</sub><sup>3-</sup>-P, PHA, glycogen in PAOs-Test, and variations of COD, and NO<sub>3</sub><sup>-</sup>-N in Denitrifiers-Test <sup>a</sup>

Treatment time	PAOs-Test			Denitrifiers-Test	
	Soluble COD (mg/L)	PO <sub>4</sub> <sup>3-</sup> -P (mg/L)	PHA (mmol C/g VSS)	Soluble COD (mg/L)	NO <sub>3</sub> <sup>-</sup> -N (mg/L)
0	242 ± 10	0.1 ± 0.1	0.56 ± 0.12	242 ± 10	35 ± 0.1
90 min	163 ± 8	13.8 ± 0.4	0.84 ± 0.15	135 ± 6	Non-detectable

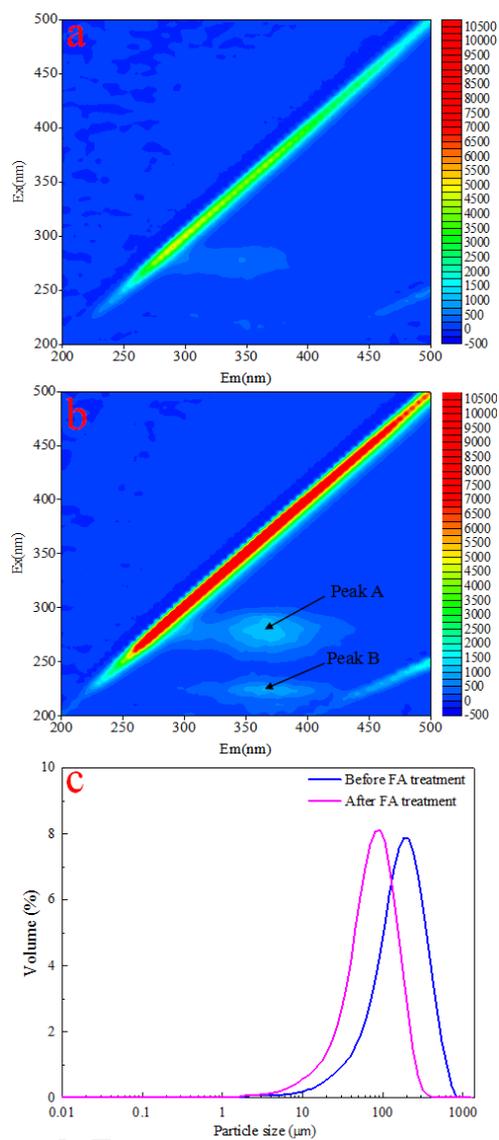
<sup>a</sup> Results are the averages and standard deviations from triplicate tests.



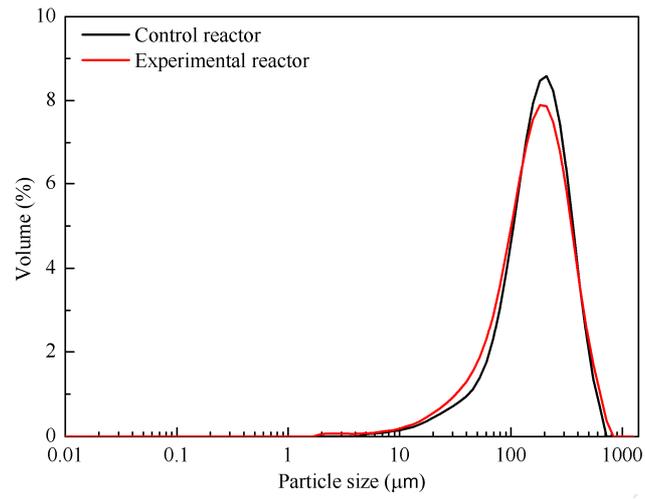
**Fig. 1.** Schematic diagram of the control (A) and experimental systems (B).



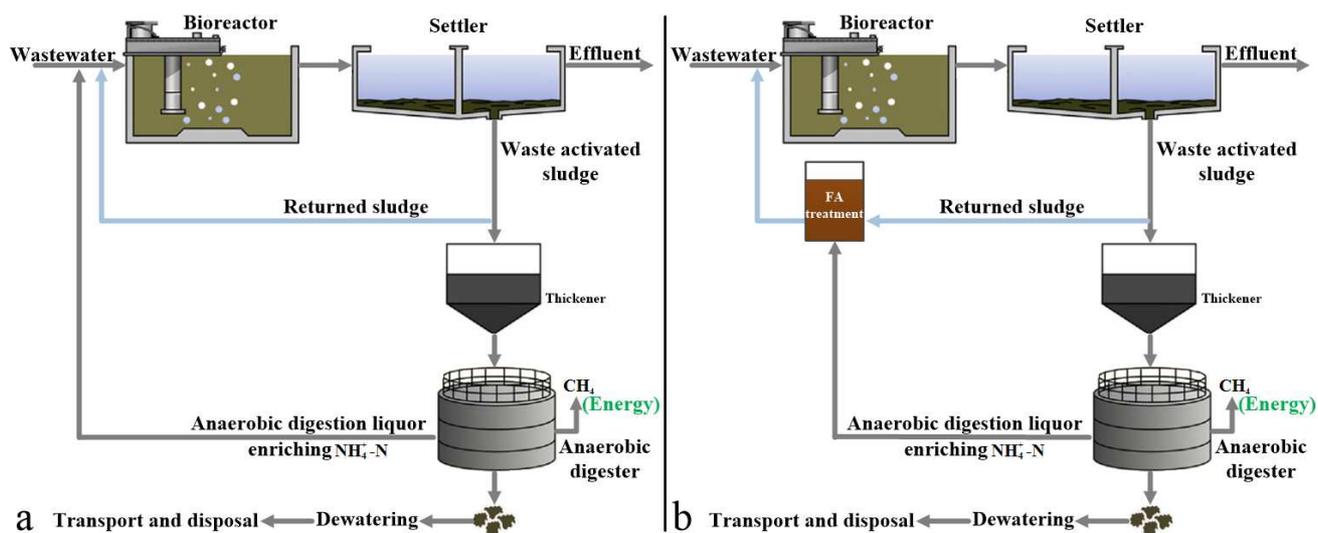
**Fig. 2.** Long-term MLSS and MLVSS concentrations in the control and experimental reactors. Phase I: before FA treatment (baseline phase); Phase II: FA treatment implemented in the experimental reactor (experimental phase).



**Fig. 3.** EEM profiles of liquid (a: before treatment; b: after treatment) and particle size distribution of sludge (c) before and after FA treatment in the FA treatment unit.



**Fig. 4.** Particle size distribution of sludge in the control and experimental treatment reactors.



**Fig. 5.** Traditional wastewater treatment process (a), innovative “closed-loop” concept in wastewater treatment process based on the proposed FA treatment technology to achieve sludge reduction (b).

**Highlights:**

- A new strategy was developed for sludge reduction from wastewater treatment process.
- The strategy was implemented through treating part of the sludge with FA.
- 16 mg N/L FA caused ~20% of sludge reduction in the experimental reactor.
- Reactor performance and sludge properties were not affected by FA treatment.