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Assessment of N₂O emissions from a fertilised vegetable cropping soil under different plant residue management strategies using ¹⁵N tracing techniques

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26 **Abstract**

27 Combined application of plant residues and N fertilisers strongly affect soil mineral N dynamics and
28 N₂O emissions depending on the quality of the plant residues, their application methods and other
29 management strategies. We investigated the effect of combined application of two vegetable plant
30 residues (cauliflower and sweet corn) and ¹⁵N fertiliser on N dynamics and N₂O emission in a
31 glasshouse pot study. The experiment was conducted under two residue management practices (soil
32 incorporation vs surface mulching) over 98 days with growing basil (*Ocimum basilicum*) plants. We
33 also assessed the efficacy of applying the nitrification inhibitor, 3,4-dimethylpyrazole phosphate
34 (DMPP) to the plant residues, for reducing N loss and mitigating N₂O emissions. Application of
35 plant residues, both on the soil surface or into soil, resulted in net N mineralisation and increased
36 cumulative N₂O emission compared with the application of N fertiliser alone. Soil surface mulching
37 of sweet corn decreased total and residue-induced cumulative N₂O emission compared with the
38 incorporation method, while it showed opposite effect on N₂O emissions from cauliflower residue.
39 The application of DMPP with sweet corn residue reduced total, residue- and fertiliser-induced N₂O
40 emissions; however its application with cauliflower residue did not show any mitigating effect on the
41 N₂O emissions. The residue application methods and the use of DMPP did not significantly affect
42 ¹⁵N recovery by the basil plants. In contrast, soil incorporation of these residues doubled the
43 microbial immobilisation of applied ¹⁵N into soil organic matter. Linear regression analysis of N₂O
44 emission during the experimental period indicated that in the treatments without DMPP application,
45 soil NO₃⁻-N concentration was the most important factor in controlling the magnitude of N₂O
46 emissions, while the application of DMPP changed the dominant regulating factor from NO₃⁻-N to
47 NH₄⁺-N concentration.

48 **Keywords:**

49 Crop residue management, Nitrification inhibitor, 3,4-Dimethylpyrazol phosphate (DMPP), Nitrous
50 oxide (N₂O), ¹⁵N tracing

51

52 **1. Introduction**

53 Agricultural lands are the major source of anthropogenic N₂O, which has a significant role in global
54 warming and destruction of the ozone layer (IPCC, 2013). Intensive vegetable cropping systems are
55 generally characterised by heavy fertiliser N applications to maintain productivity, and consequently
56 high N₂O emissions occur through microbial nitrification and denitrification (Pang et al., 2009;

57 Rezaei Rashti et al., 2015; Scheer et al., 2014). Therefore, reduction in N₂O emissions from these
58 cropping systems could potentially make a significant contribution to the mitigation of global
59 anthropogenic N₂O emissions.

60 Agricultural activities provide nearly 4 billion metric tons of plant residues per year at the global
61 scale (Lal, 2005). Returning of these residues to cropping lands can sustain soil organic matter and
62 enhance soil fertility by increasing microbial activity and nutrient availability (Ma et al., 2010; Smith
63 et al., 1993) as well as reducing water loss and limiting weed growth. It has been reported in
64 previous studies that residue application may increase or decrease N₂O emission depending on the
65 quantity and quality (nutrient content, biochemical composition and physical features) of the applied
66 residues (Baggs et al., 2003; Chen et al., 2013; Garcia-Ruiz and Baggs, 2007; Rezaei Rashti et al.,
67 2016), while denitrification is considered as the primary source of N₂O emissions in plant residue
68 amended soils (Kong et al., 2017; Li et al., 2016). Vegetable residues can release up to 150 kg N ha⁻¹
69 through mineralization (De Neve and Hofman, 1998), and different management practices of the
70 harvested vegetable residues (such as soil incorporation or surface mulching) may affect N₂O
71 emissions differently in these cropping systems. Generally, plant residues applied as a surface mulch
72 have a slower decomposition rate than soil incorporated residues due to the greater fluctuations in
73 soil moisture content and temperature, lower availability of soil nutrients and limited contact of
74 applied residues with soil in this application method (Schomberg et al., 1994; Thonissen et al.,
75 2000). Huang et al. (2004) and Zhu et al. (2013) reported that decomposition of soil incorporated
76 plant residues provided more bioavailable carbon and nitrogen sources for soil microbial activities.
77 The enhancement of soil microbial respiration may also facilitate the development of anaerobic
78 micro-sites which favour the denitrification process.

79 Combined application of plant residues and chemical N fertilisers has been reported to be beneficial
80 in increasing N use efficiency of applied fertilisers (Mohammad et al., 2012). This may occur
81 through enhancing the microbial immobilisation of applied mineral N, in the early days after
82 application, and synchronising soil N dynamics with N demands of the cultivated crop. However,
83 studies by Garcia-Ruiz and Baggs (2007) and Gentile et al. (2008) indicated that soil incorporation of
84 plant residues in combination with N fertilisers may increase N₂O emissions. Carmo et al. (2013) and
85 Wang et al. (2016) also reported increases in N₂O emissions after surface application of sugarcane
86 residue. It has been suggested that mineral N application may increase the decomposition rate of
87 labile carbon compounds in applied plant residues (Jiang et al., 2015). The increase of soluble
88 organic carbon in the presence of high levels of soil mineral N would consequently increase N₂O
89 emissions by stimulating the denitrification process (Paul and Beauchamp, 1989, Lan et al., 2017).

90 In order to reduce N losses and increase fertiliser N use efficiency, nitrification inhibitors have been
91 introduced to agricultural soils (Boeckx et al., 2005; Di and Cameron, 2003; Pereira et al., 2010).
92 Nitrification inhibitors can delay the conversion of NH_4^+ to NO_3^- and provide more opportunities for
93 plant uptake and microbial immobilization of NH_4^+ within the soil profile. The inhibition of O_2
94 consumption by the nitrification process may also improve soil O_2 status and reduce N_2O loss
95 through denitrification (Zhu et al., 2015). The 3,4-dimethylpyrazole phosphate (DMPP) is one of the
96 most popular forms of such inhibitors, which has been widely used over the past years (Hatch et al.,
97 2005; Zerulla et al., 2001). This nitrification inhibitor is effective at low application rates of 0.5-1.5
98 kg ha^{-1} . DMPP has a low water solubility, a slow degradation rate and can reduce the risk of NO_3^-
99 leaching and N_2O emission (Li et al., 2009; Menendez et al., 2006; Zerulla et al., 2001). Menendez et
100 al. (2012) reported that the addition of DMPP to mineral N fertilisers can significantly reduce N_2O
101 emissions, but the efficacy of this nitrification inhibitor strongly depends on the environmental
102 conditions. The effect of DMPP on reducing N_2O emissions from fertilised vegetable fields has been
103 investigated recently by Pfab et al. (2012) and Scheer et al. (2014), but to our knowledge no
104 published data are currently available on the mitigating effects of DMPP in combined application of
105 vegetable residues and N fertilisers.

106 The main objectives of the present study were to: (1) monitor the dynamics of soil mineral N and
107 N_2O emissions following the application of two contrasting vegetable residues with ^{15}N -labelled
108 fertiliser, in the presence of growing plants; (2) evaluate the effect of different plant residue
109 management strategies (incorporation vs. surface mulching) on N_2O emission and N use efficiency
110 of the applied N sources; (3) determine the effects of DMPP application on reducing N losses and
111 N_2O emissions following plant residue application; (4) assess the effect of moisture fluctuations on
112 soil N dynamics and N_2O production from combined applications of vegetable residues and ^{15}N -
113 labelled fertiliser. The underlying hypotheses were: (a) Vegetable residue quality and its application
114 method would affect soil N dynamics and N_2O emissions; (b) DMPP application onto vegetable
115 residues before incorporation into soil may increase N use efficiency and consequently reduce N_2O
116 emissions.

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121 2. Materials and Methods

122 2.1. Plant materials and biochemical analysis

123 Two common crop residues in Australian sub-tropical vegetable cropping systems with substantial
124 differences in N content and chemical/biochemical characteristics, namely cauliflower (*Brassica*
125 *oleracea ver. botrytis L.*) and sweet corn (*Zea mays L.*), were selected. The plant residues were dried
126 at 60°C for two days and then cut into 2 cm pieces for application to pots and ground to <1 mm for
127 chemical and biochemical analyses (Table 1). Total carbon (TC) and nitrogen (TN) contents of plant
128 materials were determined by dry combustion using a LECO CN analyser (TruMac NO. 830-300-
129 400, USA). Lignin and cellulose contents were determined sequentially with the acid detergent pre-
130 treatment method (Wang et al., 2004). Total polyphenol contents in residue samples were determined
131 using 50% methanol extractant followed by the Folin Ciocalteu colorimetric method calibrated with
132 gallic acid (Waterman and Mole, 1994). The results are reported on an oven-dry weight basis.

133 The chemistry of each plant material was also assessed with solid state ¹³C-cross-polarization magic
134 angle spinning (CPMAS) nuclear magnetic resonance (NMR) spectroscopy using a Varian Unity
135 Inova 400 spectrometer (Varian Inc., Palo Alto, CA) operating at a frequency of 100.6 MHz. A
136 measured mass of each plant material (250 mg) was packed into a silicon nitride rotor (7 mm OD)
137 and spun at 5 kHz at the magic angle. A standard cross-polarization pulse sequence was applied with
138 single contact times of 2 ms, an acquisition time of 14 ms, and a recycle delay of 2.5 s. Chemical
139 shift values were referenced externally to hexamethylbenzene at 132.1 ppm, equivalent to
140 tetramethylsilane at 0 ppm. The ¹³C CPMAS NMR spectra were divided into seven major chemical
141 shift regions: alkyl C (-50 to 45 ppm), N-alkyl/methoxyl C (45 to 60 ppm), O-alkyl C (60 to 95
142 ppm), di-O-alkyl C (95 to 110 ppm), aryl C (110 to 145 ppm), O-aryl C (145 to 165 ppm) and
143 carbonyl C (165 to 210 ppm). The relative intensities for each region were determined by integration
144 using the NMR software package MestReNova (Version 8.1.4, Mestrelab Research S.L., 2013).

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152 Table 1: Chemical and biochemical composition of the plant materials determined by chemical
 153 methods and NMR spectroscopy

Plant Materials	Chemical analysis (mg g ⁻¹ plant material)						¹³ C NMR analysis (mg g ⁻¹ plant material)						
	TC	TN	C/N	Lignin	Cellu- lose	Poly- phenol	Alkyl	N-alkyl/ methoxyl	O- alkyl	di-O- alkyl	Aryl	O- Aryl	carbo- nyl
Cauliflower	379	34	11	36	136	7	98	46	144	24	17	8	42
Sweet corn	405	15	27	38	258	8	58	43	176	46	27	15	40

154

155 2.2. Glasshouse preparation and experimental design

156 The fresh soil used in this study was collected from the top 20 cm of a cultivated vegetable field at
 157 Gatton Horticultural Research Station (27° 32' S, 152° 19' E) in the Lockyer Valley, Queensland,
 158 Australia. The soil is classified as a Vertosol (Isbell, 2002) and comprised 35% sand, 24% silt and
 159 41% clay with an initial pH of 7.7 (1:5 water) and a water holding capacity of 530 g kg⁻¹. Total
 160 organic carbon (OC) and N were 15.4 and 1.1 g kg⁻¹, respectively.

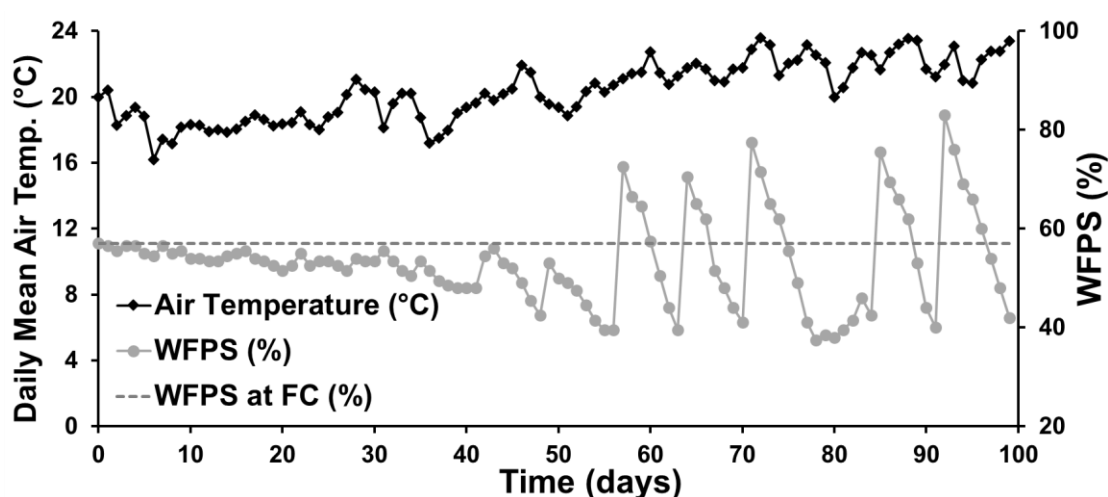
161 The chamber units specially designed for this experiment consisted of two identical cylindrical
 162 polyethylene chambers (15 cm diameter and 22 cm height). The lower part was designed as a water-
 163 tight pot and the detachable upper part was designed as a gas sampling chamber. The two parts could
 164 be connected together using an elastic rubber band (10 cm wide) and high vacuum silicon grease was
 165 applied to ensure an airtight seal between them during gas sampling.

166 The study was conducted using eight treatments (4.5 kg fresh soil, equivalent of 3.9 kg dry-mass per
 167 pot, in seven replicates) namely: (1) Control: no added N fertiliser or plant residues; (2) Urea only:
 168 50 mg N kg⁻¹ soil, equivalent to 120 kg N ha⁻¹ as 10.3 atom% ¹⁵N-urea; (3) CFI: incorporated
 169 cauliflower residue (2.08 g kg⁻¹ soil, equivalent to 5 tonnes ha⁻¹ residue in the top 20 cm of soil) +
 170 ¹⁵N-urea; (4) SCI: incorporated sweet corn residue (2.08 g kg⁻¹ soil, equivalent to 5 tonnes ha⁻¹
 171 residue in the top 20 cm of soil) + ¹⁵N-urea; (5) CFD: incorporated cauliflower residue + DMPP (0.7
 172 mg active ingredient (a.i.) kg⁻¹ soil, equivalent to 1.67 kg a.i. ha⁻¹; Incitec Pivot Ltd, Australia) + ¹⁵N-
 173 urea; (6) SCD: incorporated sweet corn residue + DMPP + ¹⁵N-urea; (7) CFS: surface applied
 174 cauliflower residue + incorporated ¹⁵N-urea; and (8) SCS: surface applied sweet corn residue +
 175 incorporated ¹⁵N-urea. The ¹⁵N-labelled urea was evenly mixed through the soil in the fertilised
 176 treatments prior to residue application. Plant residues in CFD and SCD treatments were moistened

177 with diluted liquid DMPP in sealed plastic bags for one hour and then evenly mixed with fresh soil.
 178 The soil was packed into the lower part of the chamber unit by pushing gently to a bulk density of
 179 1.2 g cm^{-3} (field bulk density). Seeds of Genovese basil (*Ocimum basilicum*) were sown after
 180 preparation of the pots. Following germination, four plants were left in each pot and plants were
 181 harvested after 98 days. The pots were moistened to 57% water-filled pore space (WFPS) with
 182 distilled water and kept close to this moisture level for the first six weeks of the study (Fig. 1). After
 183 this period (from day 42), the moisture levels in the pots were managed according to the designated
 184 fluctuating regime (40-85% WFPS) for the rest of the experiment.

185 Daily air temperature inside the glasshouse was measured using a temperature data logger (Tinytag
 186 Plus 2 TGP-4020) during the experimental period (Fig. 1). The WFPS was measured using the soil
 187 bulk density and volumetric soil moisture content of the treatments. The moisture content of each pot
 188 was adjusted by adding distilled water according to its weight loss every three days, until the end of
 189 the study.

190



191

192 Fig.1. Daily air temperature of the glasshouse, field capacity (FC) of the experimental soil (dash line)
 193 and soil moisture fluctuation in water-filled pore space (WFPS) during the study period.

194

195 2.3. Measurement of N₂O emissions

196 Gas sampling was undertaken every 1 to 4 days depending on soil moisture conditions and the
 197 expected levels of N₂O emissions. Gas samples were collected from the chambers' headspaces, in
 198 four replicates, one hour after closure using a 25 mL gas-tight syringe and immediately transferred to
 199 pre-evacuated 12 mL glass vials (Exetainer, Labco Ltd, High Wycombe, UK). The gas samples were
 200 analysed for N₂O concentration using a gas chromatograph (Varian CP-3800, Varian Inc.,

201 Middleburgh, the Netherlands) as described by Wang et al. (2011). Linearity tests on gas
202 concentration increases were performed on a subset of sampling occasions during the study for all
203 treatments by taking samples after the closure of chambers, every 30 min for 2 hours. Nitrous oxide
204 emissions showed a linear trend over the first hour of the measurement period. The emissions for
205 days without gas sampling were estimated using the arithmetic mean of the measurements on the two
206 closest days. The cumulative emissions were calculated by summing the daily emissions.

207 Fertiliser-induced N₂O emissions were calculated according to the ¹⁵N abundance in the gas samples,
208 which was detected using an automated isotope ratio mass spectrometer (IRMS) (SERCON, 20-20,
209 UK) linked to a Sercon Cryoprep trace gas preparation system. Plant residue-induced N₂O emissions
210 were estimated from total N₂O emissions by subtracting the emissions from the control and fertiliser-
211 induced emissions of each treatment.

212

213 **2.4. Soil sampling and biochemical analysis**

214 Soil samples were collected at 13, 41, 76 and 98 days after the commencement of the experiment.
215 Three replicates of each treatment were non-destructively sampled, using polyethylene columns of 2
216 cm diameter, for soil mineral N (NO₃⁻ and NH₄⁺), water soluble organic C (WSOC) and microbial
217 biomass C (MBC) on all sampling days. The NO₃⁻ and NH₄⁺ concentrations were determined by
218 colorimetric techniques (Rayment and Lyons, 2011) after extracting the fresh soil samples with 2 M
219 KCl at 1:5 ratio of soil to extractant, using a continuous segmented flow auto-analyser (SEAL
220 Analytical Quattro). The WSOC was extracted with distilled water at 1:2 ratio of soil to water and
221 determined using the colorimetric method described by Burford and Bremner (1975). The MBC was
222 determined using the chloroform fumigation-extraction method (Vance et al., 1987).

223

224 **2.5. The ¹⁵N abundance analysis of soil, plant and extracted solution**

225 The ¹⁵N abundance of NH₄⁺ and NO₃⁻ in the KCl extracts of soil was measured using a diffusion
226 technique followed by direct combustion spectrometry as described by Stark and Hart (1996) and
227 Bedard-Haughn et al. (2004). Briefly, a 5.00 to 40.00 mL aliquot (equal to 100 µg N) of 2 M KCl
228 extract was pipetted into a specimen container. After the addition of 0.2 g MgO powder, an acidified
229 filter paper (7 mm diameter Whatman No. 3 disc acidified with 10 µL of 2.5 M KHSO₄) was hung
230 inside the container using a stainless steel wire. The container was closed immediately and placed at

231 room temperature for 6 days to complete the diffusion of $^{15}\text{N-NH}_4^+$ onto the filter paper. After taking
232 the filter paper out, the container was left open overnight. At this stage, after addition of 0.4 g
233 Devarda's alloy powder, a new acidified filter paper was hung inside the container for another 6 days
234 to complete the diffusion of $^{15}\text{N-NO}_3^-$. The dried filter paper was then pushed off the wire into a
235 widened 5×8 mm Sn capsule and kept in a microtiter plate before direct combustion. To prepare
236 plant and soil samples for ^{15}N abundance analysis at the end of the experiment, harvested basil plants
237 and the soil samples after KCl extraction and repeated washing with 0.01 M CaCl_2 solution, were
238 dried at 60°C for two days, finely ground (< 150µm) and transferred to 5×8 mm Sn capsules. The
239 ^{15}N abundance in the filter papers, soil and plant tissue samples, were determined with direct
240 combustion using an Isotope Ratio Mass Spectrometer (Sercon Hydra 20-22, Sercon Europa EA-
241 GSL).

242

243 **2.6. Statistical analysis**

244 All data were statistically analysed by univariate analysis of variance using the IBM SPSS Statistics
245 23 software package. The differences at $P \leq 0.05$ between treatments using LSD test were considered
246 statistically significant and variables were tested for normality of distribution using Kolmogorov-
247 Smirnov test. Stepwise multiple linear regression analysis was used to identify relationships between
248 daily N_2O emission and different properties of the treatments.

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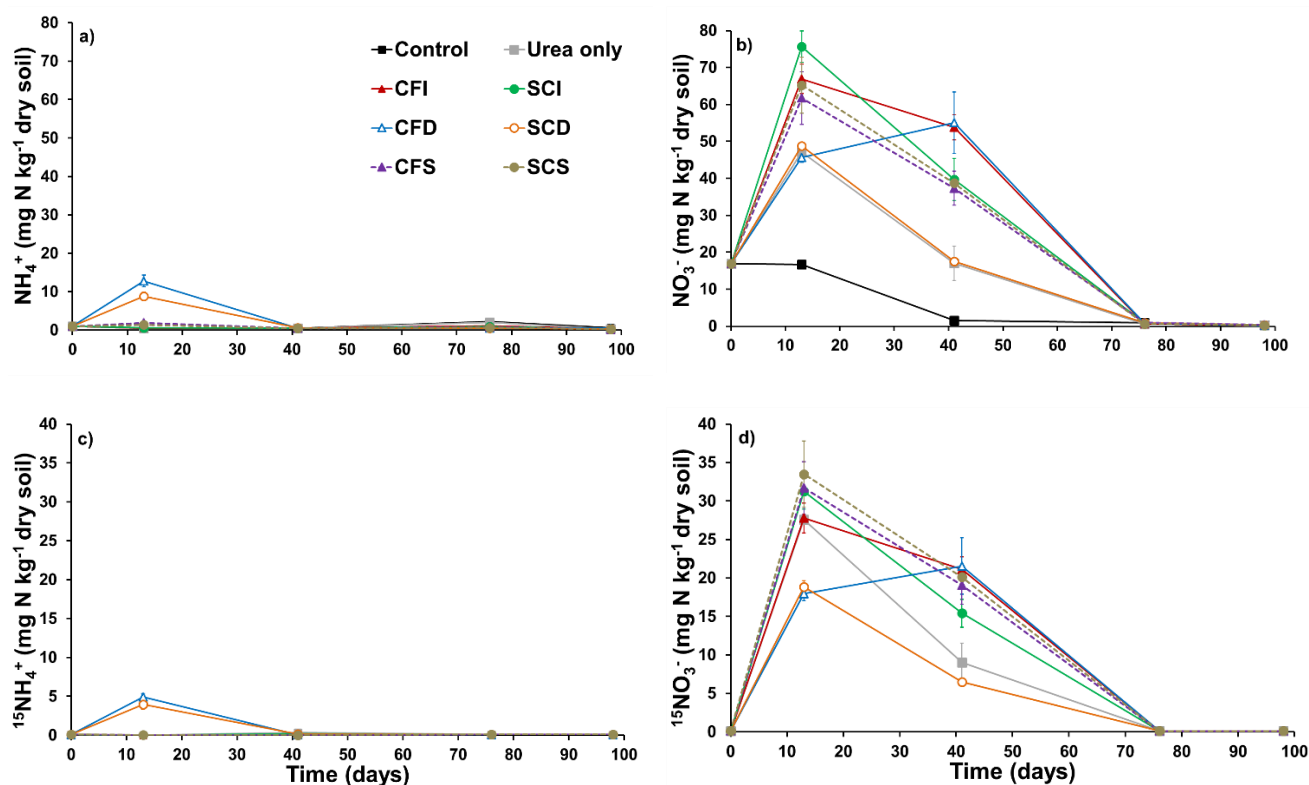
250 **3. Results**

251 **3.1. Soil mineral N dynamics**

252 The combined application of N fertiliser and plant residues in all treatments (except SCD) resulted in
253 significantly ($P < 0.05$) higher mineral N ($\text{NH}_4^+ + \text{NO}_3^-$) contents compared with the Urea only
254 treatment in the first 41 days of the experiment (Fig. 2). Generally, soil NO_3^- concentration increased
255 shortly after N fertiliser and residue applications, with the highest concentration observed around two
256 weeks after the start of experiment, and then gradually decreased throughout the plant growing
257 period. There were no significant differences in soil NH_4^+ and NO_3^- concentrations between different
258 residues (sweet corn vs cauliflower). However, surface mulching of both plant residues in the CFS
259 and SCS treatments significantly ($P < 0.05$) decreased NO_3^- concentration when compared with
260 incorporated plant materials in the CFI and SCI treatments in the early stage of decomposition.

261 The application of DMPP to plant residues in CFD and SCD treatments significantly ($P < 0.05$)
 262 reduced soil NO_3^- and fertiliser-induced $^{15}\text{NO}_3^-$ concentrations in the first two weeks of the
 263 experiment, compared with other plant material amended treatments. Although the Urea only
 264 treatment showed high $^{15}\text{NO}_3^-$ concentration at the start of the experiment, after 40 days all
 265 treatments (except SCD) showed significantly ($P < 0.05$) higher $^{15}\text{NO}_3^-$ concentration than the Urea
 266 only treatment. The amendment of DMPP significantly ($P < 0.01$) increased soil NH_4^+ and $^{15}\text{NH}_4^+$
 267 concentrations in the early days of the study when compared with other treatments. The similar
 268 concentration of fertiliser $^{15}\text{NH}_4^+$ in the presence of cauliflower and sweet corn residues implies that
 269 the residue type did not have a significant effect on DMPP efficiency in inhibiting the nitrification of
 270 applied N fertiliser.

271



272 Fig.2. Soil mineral N (a and b) and fertiliser-derived mineral ^{15}N (c and d) concentrations. Vertical
 273 bars are standard error of three replicates. CFI = Incorporated cauliflower residue + ^{15}N -urea; SCI =
 274 Incorporated sweet corn residue + ^{15}N -urea; CFD = Incorporated cauliflower residue + DMPP + ^{15}N -
 275 urea; SCD = Incorporated sweet corn residue + DMPP + ^{15}N -urea; CFS = surface applied cauliflower
 276 residue + ^{15}N -urea; SCS = surface applied sweet corn residue + ^{15}N -urea.

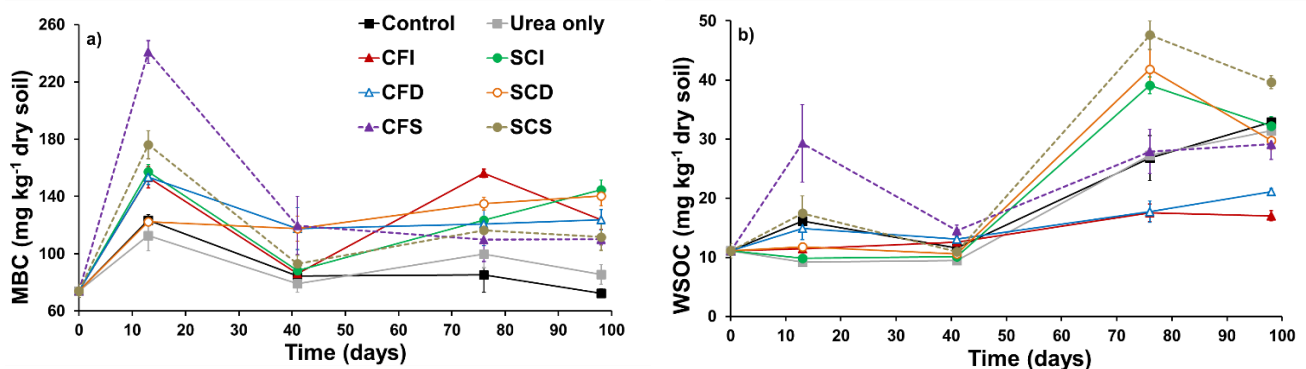
277

278 **3.2. Microbial biomass carbon and water soluble organic carbon dynamics**

279 The combined application of N fertiliser and plant residues generally increased MBC concentration
 280 compared with the control and Urea only treatments (Fig. 3).The MBC concentrations in cauliflower
 281 amended treatments were generally higher than sweet corn treatments during the experimental
 282 period, however the differences were not always statistically significant. Surface mulching of
 283 residues showed the highest MBC concentrations in the early stage of the study, however the MBC
 284 concentrations in CFS and SCS treatments then remained at moderate and relatively steady levels
 285 during the rest of the experiment. In contrast, the soil incorporated treatments showed higher MBC
 286 concentrations in the second stage of the experiment along with decomposition of applied plant
 287 residues.

288 The surface mulching treatments of both applied residues increased soil WSOC concentration in
 289 comparison with the soil incorporated treatments. Generally, the differences in WSOC
 290 concentrations between cauliflower and sweet corn treatments were not statistically significant in the
 291 early stages of the experiment; however all treatments with sweet corn amendment (SCI, SCD and
 292 SCS) had significantly ($P < 0.05$) higher WSOC compared with cauliflower amended treatments
 293 (CFI, CFD and CFS) in the second part of the study. The similar dynamic patterns of MBC and
 294 WSOC between the Control and Urea only treatments also indicated that application of N fertiliser
 295 alone, without amendment of external organic material, did not significantly increase soil microbial
 296 activity.

297



298 Fig.3. MBC (a) and WSOC (b) concentrations during the experimental period. Vertical bars are
 299 standard error of three replicates. CFI = Incorporated cauliflower residue + ¹⁵N-urea; SCI =
 300 Incorporated sweet corn residue + ¹⁵N-urea; CFD = Incorporated cauliflower residue + DMPP + ¹⁵N-

301 urea; SCD = Incorporated sweet corn residue + DMPP + ^{15}N -urea; CFS = surface applied cauliflower
302 residue + ^{15}N -urea; SCS = surface applied sweet corn residue + ^{15}N -urea.

303

304 **3.3. Daily and cumulative N₂O emissions**

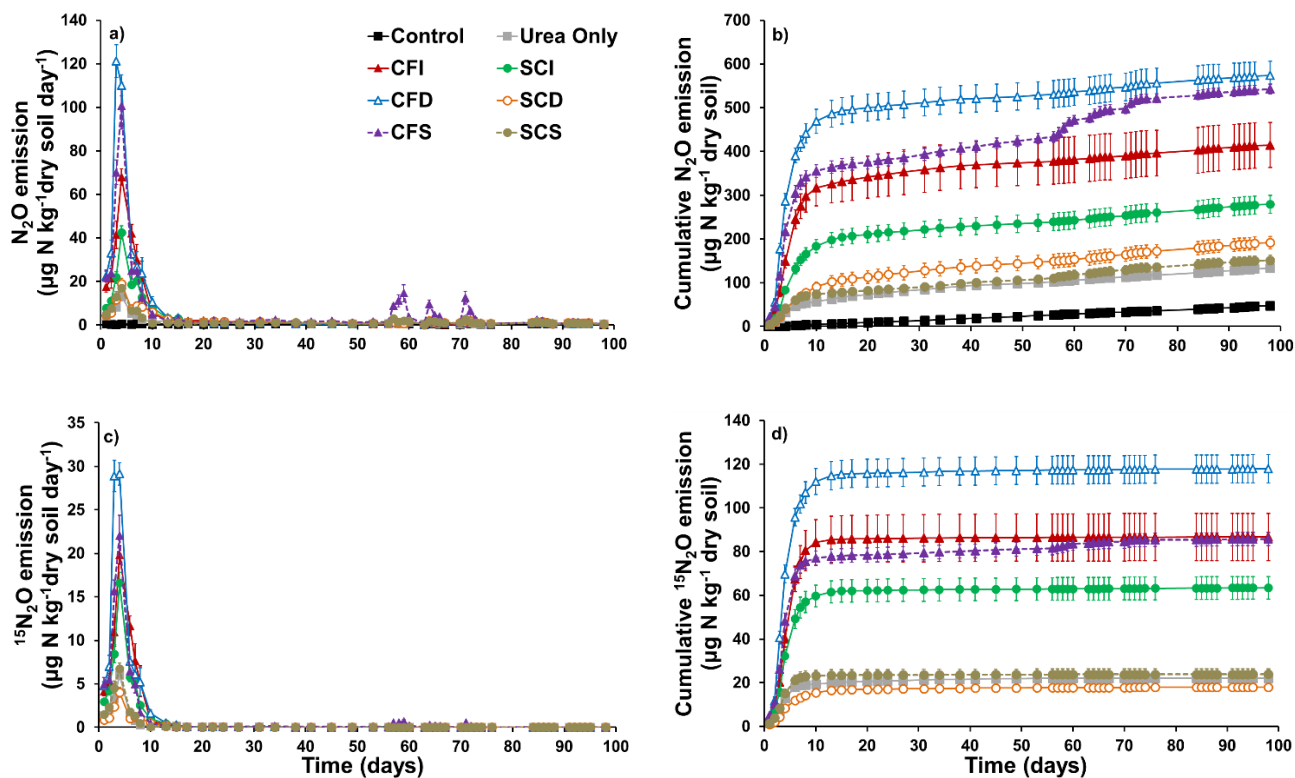
305 The N₂O emissions from all treatments, except the control, reached their highest rate within the first
306 week of the experiment (Fig. 4). The maximum daily N₂O emissions for cauliflower and sweet corn
307 residue applications were observed in the CFD (121.2 $\mu\text{g N}_2\text{O-N kg}^{-1} \text{d}^{-1}$) and SCI (42.5 $\mu\text{g N}_2\text{O-N}$
308 $\text{kg}^{-1} \text{d}^{-1}$) treatments, respectively. Generally, in all treatments except CFS, no substantial emissions
309 occurred after the first two weeks of the experiment. Therefore, moisture fluctuations in the second
310 part of the study only significantly ($P < 0.05$) increased N₂O emissions from CFS treatment.

311 Plant residue application in all treatments, except the SCS, significantly ($P < 0.05$) increased
312 cumulative N₂O emissions when compared with the Urea only treatment (Table 2). In treatments
313 with sweet corn residue, the highest cumulative N₂O emission was observed in the soil incorporated
314 condition (SCI), while the lowest N₂O emission occurred in the surface mulching of the residue
315 (SCS). The DMPP amendment to sweet corn residue (SCD) significantly ($P < 0.05$) reduced both
316 total and residue-induced N₂O emissions compared with its incorporation only method, but was not
317 able to successfully reduce N₂O emissions to levels lower than the sweet corn surface mulching and
318 Urea only treatments. In contrast, soil incorporation of cauliflower residue (CFI) resulted in the
319 lowest cumulative N₂O emission compared with CFD and CFS treatments. However, the difference
320 between cumulative N₂O emissions in CFI and CFS treatments was not statistically significant in the
321 first phase of the experiment (41 days). The application of DMPP to cauliflower residue (CFD)
322 significantly ($P < 0.05$) increased both total and residue-induced cumulative N₂O emissions compared
323 with CFI treatment.

324 The application of cauliflower and sweet corn residues to the investigated vegetable soil had
325 different effects on fertiliser-induced N₂O emissions (Fig. 4d). Soil incorporation and surface
326 mulching of cauliflower residue showed a similar effect on fertiliser-induced N₂O emission, and both
327 treatments emitted significantly ($P < 0.05$) lower fertiliser-induced N₂O than the CFD treatment. In
328 contrast, incorporation of sweet corn residue resulted in higher fertiliser-induced N₂O emission
329 compared with the SCS and SCD treatments. The surface mulching treatment and DMPP amendment
330 to sweet corn residue showed a similar pattern of fertiliser-induced N₂O emission to the Urea only
331 treatment. Thus, spraying DMPP onto the plant residues before their incorporation to soil did not
332 have a consistent effect on fertiliser-induced N₂O emission. Although DMPP amendment

333 significantly ($P < 0.05$) reduced fertiliser-induced N_2O emission in incorporated sweet corn residue
 334 treatment, to the levels close to the Urea only treatment, its application greatly boosted fertiliser-
 335 induced N_2O emission in cauliflower residue incorporated vegetable soil.

336



337 Fig.4. Daily (a and c) and cumulative (b and d) N_2O emission from different treatments and the
 338 contribution of ^{15}N fertiliser during the experimental period. Vertical bars are standard error of four
 339 replicates. CFI = Incorporated cauliflower residue + ^{15}N -urea; SCI = Incorporated sweet corn residue
 340 + ^{15}N -urea; CFD = Incorporated cauliflower residue + DMPP + ^{15}N -urea; SCD = Incorporated sweet
 341 corn residue + DMPP + ^{15}N -urea; CFS = surface applied cauliflower residue + ^{15}N -urea; SCS =
 342 surface applied sweet corn residue + ^{15}N -urea.

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350 Table 2: Cumulative N₂O emissions ($\mu\text{g N}_2\text{O-N kg}^{-1}$ dry soil) of treatments during different periods
 351 of the experiment

Treatment	Time periods								
	0 - 41 days			42 - 98 days			0 - 98 days		
	Total	Fertiliser	Crop residue	Total	Fertiliser	Crop residue	Total	Fertiliser	Crop residue
Control	18.8(f)*	-----**	-----	28.9(c)	-----	-----	47.7(f)	-----	-----
Urea Only	94.0(e)	21.7(d)	-----	39.2(b)	0.4(b)	-----	133.2(e)	22.1(d)	-----
CFI	369.6(b)	86.3 (b)	211.0(b)	45.1(b)	0.4(b)	6.0(b)	414.7(b)	86.7(b)	217.0(b)
CFD	521.2(a)	116.9(a)	332.0 (a)	52.4(b)	1.0(b)	12.7(b)	573.6(a)	117.9(a)	344.7(a)
CFS	411.5 (b)	80.4(b)	258.8 (b)	131.1(a)	5.4(a)	87.0(a)	542.6(a)	85.8(b)	345.8(a)
SCI	229.7 (c)	62.7(c)	94.8(c)	49.7(b)	0.8(b)	10.2(b)	279.5(c)	63.4(c)	105.0(c)
SCD	138.4 (d)	17.5 (d)	48.6(d)	53.1(b)	0.5(b)	13.8(b)	191.5(d)	18.0(d)	62.4(d)
SCS	100.6 (e)	23.7(d)	4.6 (e)	50.9(b)	0.2(b)	11.9(b)	151.5(e)	23.9(d)	16.5(e)

352 * The different letters in parentheses within a column indicate significant differences between the
 353 treatments ($P < 0.05$); ** Not determined. CFI = Incorporated cauliflower residue + ¹⁵N-urea; SCI =
 354 Incorporated sweet corn residue + ¹⁵N-urea; CFD = Incorporated cauliflower residue + DMPP + ¹⁵N-
 355 urea; SCD = Incorporated sweet corn residue + DMPP + ¹⁵N-urea; CFS = surface applied cauliflower
 356 residue + ¹⁵N-urea; SCS = surface applied sweet corn residue + ¹⁵N-urea.

357

358 3.4. Nitrous oxide emission in relation to soil mineral N, WSOC and MBC

359 Linear regression analysis using soil moisture content (WFPS), air temperature, WSOC, MBC, NO₃⁻-
 360 N and/or NH₄⁺-N concentrations as independent variables accounted for 45-99% of the variability in
 361 daily N₂O emissions across different treatments (Table 3). The results indicated that daily variations
 362 in N₂O emissions from the control treatment were negatively correlated with soil WSOC
 363 concentration. In Urea only and residue applications without DMPP, soil NO₃⁻-N concentration was

364 the most important factor controlling the magnitude of N₂O emissions. The application of DMPP in
 365 CFD and SCD treatments changed the dominant regulating factor of N₂O emissions from soil NO₃⁻-
 366 N to NH₄⁺-N concentrations.

367

368 Table 3: Regression equations between daily N₂O emission and treatment properties (n=12)

Treatment	Equation	R ²	P
Control	$Y^* = 0.65 - 0.01$ (WSOC)**	0.51	≤ 0.01
Urea Only	$Y = 0.34 + 0.04$ (NO ₃ ⁻)	0.82	≤ 0.01
CFI	$Y = -3.32 + 0.01$ (NO ₃ ⁻) + 0.08 (WFPS)	0.92	≤ 0.01
CFD	$Y = 0.56 + 0.25$ (NH ₄ ⁺)	0.87	≤ 0.01
CFS	$Y = 0.58 + 0.02$ (NO ₃ ⁻)	0.87	≤ 0.01
SCI	$Y = 0.29 + 0.04$ (NO ₃ ⁻)	0.51	≤ 0.01
SCD	$Y = -6.63 + 0.07$ (NH ₄ ⁺) + 0.15 (WFPS)	0.99	≤ 0.01
SCS	$Y = 0.46 + 0.01$ (NO ₃ ⁻)	0.45	≤ 0.05

369 * Daily N₂O emission (μg N₂O-N kg⁻¹ dry soil d⁻¹); ** NO₃⁻ (mg NO₃⁻-N kg⁻¹ dry soil); NH₄⁺ (mg
 370 NH₄⁺-N kg⁻¹ dry soil); WSOC (mg C kg⁻¹ dry soil); MBC (mg C kg⁻¹ dry soil). CFI = Incorporated
 371 cauliflower residue + ¹⁵N-urea; SCI = Incorporated sweet corn residue + ¹⁵N-urea; CFD =
 372 Incorporated cauliflower residue + DMPP + ¹⁵N-urea; SCD = Incorporated sweet corn residue +
 373 DMPP + ¹⁵N-urea; CFS = surface applied cauliflower residue + ¹⁵N-urea; SCS = surface applied
 374 sweet corn residue + ¹⁵N-urea.

375

376 3.5. The ¹⁵N fertiliser recovery in soil and plants

377 The application of crop residues by soil incorporation or surface mulching methods generally
 378 decreased ¹⁵N recovery by the growing plants (Table 4) compared with the Urea only treatment,
 379 although the differences were not always statistically significant. The DMPP amendment to CFD and
 380 SCD treatments showed no significant (P > 0.05) effect on ¹⁵N recovery by basil plants at the end of
 381 the study, compared to the incorporation without DMPP or surface mulching treatments. In contrast,
 382 soil incorporation of plant residues almost doubled ¹⁵N recovery in soil organic N and the application
 383 of DMPP further promoted this process by around 5% for both applied residues. In addition, surface

384 mulching of residues also significantly ($P < 0.05$) increased ^{15}N recovery in soil compared with Urea
 385 only treatment.

386 Plant residue application generally increased fertiliser-induced N_2O emissions (except in SCS and
 387 SCD) compared with the Urea only treatment. The results also indicated that DMPP application only
 388 reduced fertiliser-induced N_2O emissions when applied to sweet corn residue, while its application to
 389 cauliflower residue showed the highest fertiliser-induced N_2O emissions among all applied
 390 treatments. Surface mulching of plant residues did not have a significant effect on gaseous ^{15}N losses
 391 compared with the Urea only treatment, while soil incorporation of sweet corn and cauliflower
 392 residues significantly ($P < 0.05$) reduced total gaseous ^{15}N losses during the study period. The
 393 treatments with DMPP amendment also further reduced N fertiliser loss by around 3% and 12%
 394 compared with plant residue incorporation without DMPP and surface mulching treatments,
 395 respectively.

396 Table 4: The ^{15}N fertiliser fate in soil, air and plant tissues

Treatment	Fertiliser N uptake by plant (%)	Fertiliser N residue in soil (%)	Fertiliser N loss as N_2O (%)	Non- N_2O fertiliser N loss (%)*	Total ^{15}N (%)
Urea Only	60.06 (a)	14.64 (e)	0.04 (d)	25.26 (a)	100
CFI	56.32 (ab)	27.20 (c)	0.17 (b)	16.31 (bc)	100
CFD	51.92 (b)	32.88 (a)	0.23 (a)	14.97 (bc)	100
CFS	57.03 (ab)	18.22 (d)	0.17 (b)	24.58 (a)	100
SCI	54.04 (b)	26.82 (c)	0.12 (c)	19.02 (b)	100
SCD	55.77 (ab)	30.49 (b)	0.04 (d)	13.70 (c)	100
SCS	52.94 (b)	18.52 (d)	0.05 (d)	28.49 (a)	100

397 * The reported values were not measured directly but estimated by differences between the total
 398 amount added and the other measured sources. CFI = Incorporated cauliflower residue + ^{15}N -urea;
 399 SCI = Incorporated sweet corn residue + ^{15}N -urea; CFD = Incorporated cauliflower residue + DMPP
 400 + ^{15}N -urea; SCD = Incorporated sweet corn residue + DMPP + ^{15}N -urea; CFS = surface applied
 401 cauliflower residue + ^{15}N -urea; SCS = surface applied sweet corn residue + ^{15}N -urea.

402
 403
 404

405 **4. Discussion**

406 **4.1. Effect of soil moisture fluctuation on N dynamics and N₂O emission**

407 Many studies have reported that soil moisture content significantly affects the activity of nitrifiers
408 and denitrifiers, by altering the availability of substrates and the rate of exchange of oxygen between
409 soil and the ambient atmosphere (Ludwig et al., 2001; Petersen et al., 2008; Skiba and Smith, 2000).
410 While it was expected that soil moisture fluctuation in the second phase of the current experiment
411 (from day 42 to 98 of the study) would increase mineral N concentration due to stimulating the
412 decomposition process of applied crop residues, such an increase did not materialize in any of the
413 treatments. This observation may be attributed to the balance between N mineralisation and plant N
414 uptake during the late stages of the cropping season. This behaviour may control mineral N
415 bioavailability for microbial activities and consequently regulate N dynamics and N₂O emissions
416 after combined application of mineral N fertiliser and plant residues.

417 It has been stated by Sposito (1989) that at high soil moisture conditions, the reduction of soil redox
418 potential to optimum levels for the denitrification process and N₂O emissions may only take a few
419 hours. Soil moisture increase/fluctuation in the second phase of the current experiment increased
420 WSOC in all of the treatments but the low concentration and bioavailability of soil mineral N, rather
421 than labile C availability, for microbial activities resulted in insignificant changes in N₂O emission
422 rates (except CFS) in that period of time. The result implies that mineral N limitation, rather than the
423 availability of C sources for microbial activity, probably restricted N₂O emission from the wet soil.
424 Due to the relatively low contribution of ¹⁵N fertiliser in the emitted N₂O after increasing soil
425 moisture contents, the increased N₂O emission in the CFS treatment in this time period would be
426 regulated by the N mineralization process through decomposition of the applied cauliflower residue.
427 The large residue-induced N₂O emission peaks from the CFS, but not SCS, treatment during the late
428 wet and moisture fluctuating period, also imply that surface mulching of high N content plant
429 residues may significantly increase the N₂O emissions after heavy irrigation or rainfall events in the
430 late cropping season. This loss process may take a longer time after application of low N content
431 plant residues and would probably occur in the fallow period after harvest of the cultivated crop.

432

433

434

435 **4.2. Effect of different plant residue application and placement on N₂O emission and N** 436 **recovery**

437 The applied plant residues resulted in different N₂O emission patterns after their placement in the
438 soil. Different application methods of the high N content cauliflower residue did not show any
439 significant effect on total N₂O emissions during the first 40 days of the experiment, which is
440 consistent with the findings of Nett et al. (2015) who reported similar cumulative N₂O emissions
441 after soil incorporation and surface application of cauliflower residue to a sandy soil. However,
442 surface mulching of cauliflower residue significantly ($P < 0.05$) increased the N₂O emissions,
443 compared with its incorporation method, during the second phase of the study. These observations
444 may reflect the higher activity of soil denitrifying microbial community during decomposition of this
445 plant residue under surface mulching condition and release of more organic C and N compounds in
446 this treatment during the second phase of the experiment. Although surface mulching of cauliflower
447 residue did not increase fertiliser-induced N₂O emissions, this application method enhanced the
448 residue-induced N₂O emissions by 37%, compared with the soil incorporation method, at the end of
449 the experiment. It can be concluded that the combined application of N fertilisers with high N
450 content plant residues, either incorporated or surface mulched, should be avoided during the
451 following cropping season.

452 In contrast, surface mulching of the low N content sweet corn residue reduced plant residue-induced
453 N₂O emissions, compared with its soil incorporation, in the first 40 days of the experiment; however
454 its effect on the plant residue-induced and total N₂O emissions was small and insignificant
455 afterwards. The lower N₂O emission could be attributed to the effect of this application method on
456 increasing soil WSOC in the early stage of the experiment and potentially increasing the activity of
457 soil microbial community to immobilise more mineral N, which may consequently reduce the risk of
458 large N₂O emissions. Flessa et al. (2002) also suggested that the different decomposition and
459 denitrification behaviours of surface applied plant residues maybe partially regulated by their
460 specific 'indigenous' microflora. By the end of the experiment, the SCS treatment significantly ($P <$
461 0.05) decreased residue- and fertiliser-induced N₂O emissions by 84% and 62% compared with the
462 SCI treatment, respectively. Therefore, surface mulching of low N content plant residues seems to be
463 a promising method of reducing N₂O emission by minimising their direct contact with applied
464 mineral N fertilisers. Based on these findings, surface mulching of low N content crop residues in
465 combination with mineral N fertiliser applications may be recommended in vegetable fields under
466 intensive irrigation systems, or in high rainfall zones, in order to reduce N₂O emissions.

467 The findings of this study also suggest that the incorporated plant residues promote immobilisation
468 of available mineral N from applied N fertiliser, which is evident from significantly higher
469 recoveries of applied N fertiliser in the soil organic matter. In contrast, surface mulching of both
470 plant residues reduced the N recovery of the applied N fertiliser in soil and plant tissues by around
471 9%, compared with their soil incorporation method. Therefore it can be concluded that surface
472 mulching of plant residues significantly ($P < 0.05$) reduced the immobilisation of applied fertiliser N
473 by soil microorganisms.

474

475 **4.3. Influence of nitrification inhibitor (DMPP) on N dynamics and N₂O emission**

476 The amendment of DMPP successfully reduced total and fertiliser-derived NO₃⁻ with both applied
477 plant residues in the first two weeks of the experiment. After this initial period, DMPP gradually lost
478 its inhibition effect on the nitrification process as evidenced from the significant ($P < 0.05$) increase
479 of the total and fertiliser-derived NO₃⁻ in the cauliflower incorporated treatment between days 13 to
480 41 of the study. Both applied crop residues showed almost the same magnitudes of total fertiliser-
481 and residue-induced N₂O emissions in this period of time, which is similar to their single
482 incorporation without DMPP application. The short persistence time and low efficacy of DMPP in
483 this experiment could be attributed to its fast biological decomposition (Zerulla et al., 2001) due to
484 the high temperature and low soil water content in the early days of the experiment following its
485 application. This is consistent with the findings of Irigoyen et al. (2003), who reported the half-lives
486 of 18 and 8 days for NH₄⁺ concentration after soil incorporation of DMPP coated fertiliser at 20 °C
487 and 30 °C, respectively. The performance of DMPP in reducing N₂O emission has also been reported
488 to be more efficient in cold and wet conditions (Menendez et al., 2012; Merino et al., 2005).

489 Despite the significant effect of DMPP in inhibiting the nitrification process, its application did not
490 show a consistent effect on N₂O emissions. Although DMPP significantly ($P < 0.05$) reduced N₂O
491 emissions in the SCD treatment during high emission peaks, its application significantly ($P < 0.05$)
492 increased N₂O emissions in the CFD treatment. The increasing effect of DMPP amendment on N₂O
493 emissions from the cauliflower applied treatment may be attributed to the acidic nature of its
494 saturated solution (ca. pH = 3) which may locally reduce soil pH and consequently increase the
495 N₂O/N₂ production ratio of denitrification (Liu et al., 2010). It may also be partly related to the low
496 efficiency of the DMPP in clay soils due to its positive charge and high sorption capacity to soil clay
497 fractions (Barth et al., 2001) and insufficient concentration of applied DMPP (0.7 mg a.i. kg⁻¹ soil) to
498 this high N content plant residue. The findings of this study suggest that the DMPP's high efficiency

499 for inhibiting the nitrification process does not always coincide with its efficiency in reducing N₂O
500 emissions from applied plant residues. However, since the effect of DMPP on N mineralisation is
501 still unclear, it is hard to predict its efficacy for reducing N₂O emission from plant residues with
502 different biochemical characteristics.

503

504 **5. Conclusion**

505 In the present study, the application of high N content cauliflower residue, either incorporated into
506 soil or surface mulched, in combination with N fertiliser increased N₂O emission compared with the
507 Urea only treatment. The low N content sweet corn residue also increased N₂O emission when
508 incorporated with N fertiliser, while it significantly mitigated N₂O emission when mulched on the
509 surface. The amendment of DMPP did not show a consistent effect on N₂O emissions, suggesting
510 that the high efficacy of DMPP for inhibiting the nitrification process does not always coincide with
511 reduction in N₂O emissions. Application of DMPP with sweet corn residue decreased N₂O emission,
512 but significantly increased N₂O emission when applied with cauliflower residue. However, higher
513 concentrations of DMPP may be more efficient in reducing plant residue- and fertiliser-induced N₂O
514 emissions from high N content plant residues. Investigating the effect of plant residues of varying
515 biochemical characteristics and assessing other fertiliser and plant residue management strategies
516 may also improve understanding of these complex systems. The use of DMPP combined with
517 different plant residue application methods needs to be evaluated against its benefits in increasing N
518 use efficiency and crop yield to determine whether the recommended management strategy can be
519 affordably adopted or remains cost-prohibitive.

520

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527

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529 **References**

- 530 Baggs, E.M., Stevenson, M., Pihlatie, M., Regar, A., Cook, H., Cadisch, G., 2003. Nitrous oxide
 531 emissions following application of residues and fertiliser under zero and conventional tillage.
 532 *Plant Soil* 254, 361–370. doi:10.1023/a:1025593121839
- 533 Barth, G., Tucher, S. von, Schmidhalter, U., 2001. Influence of soil parameters on the effect of 3,4-
 534 dimethylpyrazole-phosphate as a nitrification inhibitor. *Biol. Fertil. Soils* 34, 98–102.
 535 doi:10.1007/s003740100382
- 536 Bedard-Haughn, A., Tate, K.W., van Kessel, C., 2004. Using Nitrogen-15 to Quantify Vegetative
 537 Buffer Effectiveness for Sequestering Nitrogen in Runoff. *J. Environ. Qual.* 33, 2252–2262.
 538 doi:10.2134/jeq2004.2252
- 539 Boeckx, P., Xu, X., Van Cleemput, O., 2005. Mitigation of N₂O and CH₄ Emission from Rice and
 540 Wheat Cropping Systems Using Dicyandiamide and Hydroquinone. *Nutr. Cycl.*
 541 *Agroecosystems* 72, 41–49. doi:10.1007/s10705-004-7352-4
- 542 Burford, J.R., Bremner, J.M., 1975. Relationships between the denitrification capacities of soils and
 543 total, water-soluble and readily decomposable soil organic matter. *Soil Biol. Biochem.* 7, 389–
 544 394. doi:10.1016/0038-0717(75)90055-3
- 545 Carmo, J.B. do, Filoso, S., Zotelli, L.C., de Sousa Neto, E.R., Pitombo, L.M., Duarte-Neto, P.J.,
 546 Vargas, V.P., Andrade, C.A., Gava, G.J.C., Rossetto, R., Cantarella, H., Neto, A.E., Martinelli,
 547 L.A., 2013. Infield greenhouse gas emissions from sugarcane soils in Brazil: effects from
 548 synthetic and organic fertilizer application and crop trash accumulation. *GCB Bioenergy* 5,
 549 267–280. doi:10.1111/j.1757-1707.2012.01199.x
- 550 Chen, H., Li, X., Hu, F., Shi, W., 2013. Soil nitrous oxide emissions following crop residue addition:
 551 a meta-analysis. *Glob. Chang. Biol.* 19, 2956–2964. doi:10.1111/gcb.12274
- 552 De Neve, S., Hofman, G., 1998. N mineralization and nitrate leaching from vegetable crop residues
 553 under field conditions: a model evaluation. *Soil Biol. Biochem.* 30, 2067–2075.
 554 doi:10.1016/S0038-0717(98)00082-0
- 555 Di, H.J., Cameron, K.C., 2003. Mitigation of nitrous oxide emissions in spray-irrigated grazed
 556 grassland by treating the soil with dicyandiamide, a nitrification inhibitor. *Soil Use Manag.* 19,
 557 284–290. doi:10.1111/j.1475-2743.2003.tb00317.x
- 558 Flessa, H., Potthoff, M., Loftfield, N., 2002. Greenhouse estimates of CO₂ and N₂O emissions
 559 following surface application of grass mulch: importance of indigenous microflora of mulch.
 560 *Soil Biol. Biochem.* 34, 875–879. doi:10.1016/S0038-0717(02)00028-7
- 561 Garcia-Ruiz, R., Baggs, E., 2007. N₂O emission from soil following combined application of
 562 fertiliser-N and ground weed residues. *Plant Soil* 299, 263–274. doi:10.1007/s11104-007-9382-
 563 6
- 564 Gentile, R., Vanlauwe, B., Chivenge, P., Six, J., 2008. Interactive effects from combining fertilizer
 565 and organic residue inputs on nitrogen transformations. *Soil Biol. Biochem.* 40, 2375–2384.
 566 doi:10.1016/j.soilbio.2008.05.018
- 567 Hatch, D., Trindade, H., Cardenas, L., Carneiro, J., Hawkins, J., Scholefield, D., Chadwick, D.,
 568 2005. Laboratory study of the effects of two nitrification inhibitors on greenhouse gas emissions
 569 from a slurry-treated arable soil: impact of diurnal temperature cycle. *Biol. Fertil. Soils* 41, 225–
 570 232. doi:10.1007/s00374-005-0836-9
- 571 Huang, Y., Zou, J., Zheng, X., Wang, Y., Xu, X., 2004. Nitrous oxide emissions as influenced by
 572 amendment of plant residues with different C:N ratios. *Soil Biol. Biochem.* 36, 973–981.

- 573 doi:10.1016/j.soilbio.2004.02.009
- 574 IPCC, 2013. Climate Change 2013: the Physical Science Basis. Cambridge University Press:
575 Cambridge, UK.
- 576 Irigoyen, I., Muro, J., Azpilikueta, M., Aparicio-Tejo, P., Lamsfus, and C., Irigoyen, I., Muro, J.,
577 Azpilikueta, M., Aparicio-Tejo, P., Lamsfus, and C., 2003. Ammonium oxidation kinetics in
578 the presence of nitrification inhibitors DCD and DMPP at various temperatures. *Aust. J. Soil*
579 *Res.* 41, 1177–1183. doi:10.1071/SR02144
- 580 Isbell, R.F., 2002. The Australian soil classification, Revised. ed. CSIRO Publishing: Melbourne.
- 581 Jiang, C., Yu, W., Ma, Q., Xu, Y., Zhou, H., 2015. Nitrogen addition alters carbon and nitrogen
582 dynamics during decay of different quality residues, *Ecological Engineering*.
583 doi:10.1016/j.ecoleng.2015.04.093
- 584 Kong, X., Duan, Y., Schramm, A., Eriksen, J., Holmstrup, M., Larsen, T., Bol, R., Petersen, S.O.,
585 2017. Mitigating N₂O emissions from clover residues by 3,4-dimethylpyrazole phosphate
586 (DMPP) without adverse effects on the earthworm *Lumbricus terrestris*. *Soil Biol. Biochem.*
587 104, 95–107. doi:10.1016/j.soilbio.2016.10.012
- 588 Lal, R., 2005. World crop residues production and implications of its use as a biofuel. *Environ. Int.*
589 31, 575–584. doi:10.1016/j.envint.2004.09.005
- 590 Lan, Z.M., Chen, C.R., Rezaei Rashti, M., Yang, H., Zhang, D.K., 2017. Stoichiometric ratio of
591 dissolved organic carbon to nitrate regulates nitrous oxide emission from the biochar-amended
592 soils. *Sci. Total Environ.* 576, 559–571. doi:10.1016/j.scitotenv.2016.10.119
- 593 Li, H., Chen, Y., Liang, X., Lian, Y., Li, W., 2009. Mineral-nitrogen leaching and ammonia
594 volatilization from a rice-rapeseed system as affected by 3,4-dimethylpyrazole phosphate. *J.*
595 *Environ. Qual.* 38, 2131–2137. doi:10.2134/jeq2008.0476
- 596 Li, X., Sorensen, P., Olesen, J.E., Petersen, S.O., 2016. Evidence for denitrification as main source of
597 N₂O emission from residue-amended soil. *Soil Biol. Biochem.* 92, 153–160.
598 doi:10.1016/j.soilbio.2015.10.008
- 599 Liu, B., Morkved, P.T., Frostegard, A., Bakken, L.R., 2010. Denitrification gene pools, transcription
600 and kinetics of NO, N₂O and N₂ production as affected by soil pH. *FEMS Microbiol. Ecol.* 72,
601 407–417.
- 602 Ludwig, J., Meixner, F.X., Vogel, B., Forstner, J., 2001. Soil-air exchange of nitric oxide: An
603 overview of processes, environmental factors, and modeling studies. *Biogeochemistry* 52, 225–
604 257.
- 605 Ma, E., Zhang, G., Ma, J., Xu, H., Cai, Z., Yagi, K., 2010. Effects of rice straw returning methods on
606 N₂O emission during wheat-growing season. *Nutr. Cycl. Agroecosystems* 88, 463–469.
607 doi:10.1007/s10705-010-9369-1
- 608 Menendez, S., Barrena, I., Setien, I., Gonzalez-Murua, C., Estavillo, J., 2012. Efficiency of
609 nitrification inhibitor DMPP to reduce nitrous oxide emissions under different temperature and
610 moisture conditions. *Soil Biol. Biochem.* 53, 82–89. doi:10.1016/j.soilbio.2012.04.026
- 611 Menendez, S., Merino, P., Pinto, M., Gonzalez-Murua, C., Estavillo, J.M., 2006. 3,4-
612 Dimethylpyrazol phosphate effect on nitrous oxide, nitric oxide, ammonia, and carbon dioxide
613 emissions from grasslands. *J. Environ. Qual.* 35, 973–981. doi:10.2134/jeq2005.0320
- 614 Merino, P., Menendez, S., Pinto, M., Gonzalez-Murua, C., Estavillo, J.M., 2005. 3, 4-
615 Dimethylpyrazole phosphate reduces nitrous oxide emissions from grassland after slurry
616 application. *Soil Use Manag.* 21, 53–57. doi:10.1111/j.1475-2743.2005.tb00106.x

- 617 Mohammad, W., Shah, S.M., Shehzadi, S., Shah, S.A., 2012. Effect of tillage, rotation and crop
618 residues on wheat crop productivity, fertilizer nitrogen and water use efficiency and soil organic
619 carbon status in dry area (rainfed) of north-west Pakistan. *J. Soil Sci. Plant Nutr.* 12, 715–727.
- 620 Nett, L., Fub, R., Flessa, H., Fink, M., 2015. Emissions of nitrous oxide and ammonia from a sandy
621 soil following surface application and incorporation of cauliflower leaf residues. *J. Agric. Sci.*
622 153, 1341–1352. doi:10.1017/S0021859615000027
- 623 Pang, X., Mu, Y., Lee, X., Fang, S., Yuan, J., Huang, D., 2009. Nitric oxides and nitrous oxide
624 fluxes from typical vegetables cropland in China: Effects of canopy, soil properties and field
625 management. *Atmos. Environ.* 43, 2571–2578. doi:10.1016/j.atmosenv.2009.02.016
- 626 Paul, J.W., Beauchamp, E.G., 1989. Relationship between volatile fatty acids, total ammonia, and pH
627 in manure slurries. *Biol. Wastes* 29, 313–318. doi:10.1016/0269-7483(89)90022-0
- 628 Pereira, J., Fanguero, D., Chadwick, D.R., Misselbrook, T.H., Coutinho, J., Trindade, H., 2010.
629 Effect of cattle slurry pre-treatment by separation and addition of nitrification inhibitors on
630 gaseous emissions and N dynamics: a laboratory study. *Chemosphere* 79, 620–627.
631 doi:10.1016/j.chemosphere.2010.02.029
- 632 Petersen, S., Schjonning, P., Thomsen, I., Christensen, B., 2008. Nitrous oxide evolution from
633 structurally intact soil as influenced by tillage and soil water content. *Soil Biol. Biochem.* 40,
634 967–977. doi:10.1016/j.soilbio.2007.11.017
- 635 Pfab, H., Palmer, I., Buegger, F., Fiedler, S., Muller, T., Ruser, R., 2012. Influence of a nitrification
636 inhibitor and of placed N-fertilization on N₂O fluxes from a vegetable cropped loamy soil.
637 *Agric. Ecosyst. Environ.* 150, 91–101. doi:10.1016/j.agee.2012.01.001
- 638 Rayment, G.E., Lyons, D.J., 2011. *Soil Chemical Methods-Australia*. CSIRO Publishing,
639 Collingwood, Victoria.
- 640 Rezaei Rashti, M., Wang, W.J., Moody, P.W., Chen, C.R., Ghadiri, H., 2015. Fertiliser-induced
641 nitrous oxide emissions from vegetable production in the world and the regulating factors: A
642 review. *Atmos. Environ.* 112, 225–233. doi:10.1016/j.atmosenv.2015.04.036
- 643 Rezaei Rashti, M., Wang, W.J., Reeves, S.H., Harper, S.M., Moody, P.W., Chen, C.R., 2016.
644 Linking chemical and biochemical composition of plant materials to their effects on N₂O
645 emissions from a vegetable soil. *Soil Biol. Biochem.* 103, 502–511.
646 doi:10.1016/j.soilbio.2016.09.019
- 647 Scheer, C., Rowlings, D.W., Firrel, M., Deuter, P., Morris, S., Grace, P.R., 2014. Impact of
648 nitrification inhibitor (DMPP) on soil nitrous oxide emissions from an intensive broccoli
649 production system in sub-tropical Australia. *Soil Biol. Biochem.* 77, 243–251.
650 doi:10.1016/j.soilbio.2014.07.006
- 651 Schomberg, H.H., Steiner, J.L., Unger, P.W., 1994. Decomposition and Nitrogen Dynamics of Crop
652 Residues: Residue Quality and Water Effects. *Soil Sci. Soc. Am. J.* 58, 372–381.
653 doi:10.2136/sssaj1994.03615995005800020019x
- 654 Skiba, U., Smith, K.A., 2000. The control of nitrous oxide emissions from agricultural and natural
655 soils. *Chemosph. - Glob. Chang. Sci.* 2, 379–386.
- 656 Smith, J.L., Papendick, R.L., Bezdicek, D.F., Lynch, J.M., 1993. Soil organic matter dynamics and
657 crop residue management. In: Meeting, F. B. Jr., (Ed.), *Soil Microbial ecology: Application in*
658 *Agricultural and Environmental Management*. Marcel Dekker Inc, New York, pp. 65–94.
- 659 Sposito, G., 1989. *The chemistry of soils*. Madison: Oxford University Press.
- 660 Stark, J.M., Hart, S.C., 1996. Diffusion Technique for Preparing Salt Solutions, Kjeldahl Digests,
661 and Persulfate Digests for Nitrogen-15 Analysis. *Soil Sci. Soc. Am. J.* 60, 1846–1855.

662 doi:10.2136/sssaj1996.03615995006000060033x

663 Thonissen, C., Midmore, D.J., Ladha, J.K., Olk, D.C., Schmidhalter, U., 2000. Legume
664 Decomposition and Nitrogen Release When Applied as Green Manures to Tropical Vegetable
665 Production Systems. *Agron. J.* 92, 253–260. doi:10.2134/agronj2000.922253x

666 Vance, E.D., Brookes, P.C., Jenkinson, D.S., 1987. An extraction method for measuring soil
667 microbial biomass C. *Soil Biol. Biochem.* 19, 703–707. doi:10.1016/0038-0717(87)90052-6

668 Wang, W.J., Baldock, J.A., Dalal, R.C., Moody, P.W., 2004. Decomposition dynamics of plant
669 materials in relation to nitrogen availability and biochemistry determined by NMR and wet-
670 chemical analysis. *Soil Biol. Biochem.* 36, 2045–2058. doi:10.1016/j.soilbio.2004.05.023

671 Wang, W.J., Dalal, R.C., Reeves, S.H., Butterbach-Bahl, K., Kiese, R., 2011. Greenhouse gas fluxes
672 from an Australian subtropical cropland under long-term contrasting management regimes.
673 *Glob. Chang. Biol.* 17, 3089–3101. doi:10.1111/j.1365-2486.2011.02458.x

674 Wang, W.J., Reeves, S.H., Salter, B., Moody, P.W., Dalal, R.C., 2016. Effects of urea formulations,
675 application rates and crop residue retention on N₂O emissions from sugarcane fields in
676 Australia. *Agric. Ecosyst. Environ.* 216, 137–146. doi:10.1016/j.agee.2015.09.035

677 Waterman, P.G., Mole, S., 1994. *Analysis of Phenolic Plant Metabolites*. Blackwell Scientific,
678 Boston.

679 Zerulla, W., Barth, T., Dressel, J., Erhardt, K., Horchler von Locquenghien, K., Pasda, G., Radle, M.,
680 Wissemeier, A., 2001. 3,4-Dimethylpyrazole phosphate (DMPP) - a new nitrification inhibitor
681 for agriculture and horticulture. *Biol. Fertil. Soils* 34, 79–84. doi:10.1007/s003740100380

682 Zhu, K., Bruun, S., Larsen, M., Glud, R.N., Jensen, L.S., 2015. Heterogeneity of O₂ dynamics in soil
683 amended with animal manure and implications for greenhouse gas emissions. *Soil Biol.*
684 *Biochem.* 84, 96–106. doi:10.1016/j.soilbio.2015.02.012

685 Zhu, T., Zhang, J., Yang, W., Cai, Z., 2013. Effects of organic material amendment and water
686 content on NO, N₂O, and N₂ emissions in a nitrate-rich vegetable soil. *Biol. Fertil. Soils* 49,
687 153–163. doi:10.1007/s00374-012-0711-4

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